This article was downloaded by: On: 26 January 2011 Access details: Access Details: Free Access Publisher Taylor & Francis Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37- 41 Mortimer Street, London W1T 3JH, UK

Organic Preparations and Procedures International

Publication details, including instructions for authors and subscription information: <http://www.informaworld.com/smpp/title~content=t902189982>

HETEROSUBSTITUTED CARBOCYCLIC α-AMINO ACIDS. A REVIEW

M. L. Gelmi^{ab}; D. Pocar^{ab} a Istituto di Chimica Organica, Facoltà di Farmacia Centro Interuniversitario di Ricerca sulle Reazioni Pericicliche, Milano, ITALY ^b Sintesi di Sistemi Etero e Carbociclici, Università degli Studi di Milano, Milano, ITALY

To cite this Article Gelmi, M. L. and Pocar, D.(2003) 'HETEROSUBSTITUTED CARBOCYCLIC α -AMINO ACIDS. A REVIEW', Organic Preparations and Procedures International, 35: 2, 141 — 205 To link to this Article: DOI: 10.1080/00304940309355832 URL: <http://dx.doi.org/10.1080/00304940309355832>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use:<http://www.informaworld.com/terms-and-conditions-of-access.pdf>

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

HETEROSUBSTITUTED CARBOCYCLIC *&AMINO* **ACIDS** . **A REVIEW**

 \sim \sim

^M. **L** . **Gelmi*** and **D** . **Pocar**

Istituto di Chimica Organica. Facoltd di Farmacia Centro Interuniversitario di Ricerca sulle Reationi Pericicliche e Sintesi di Sistemi Etero e Carbociclici Universitd degli Studi di Milano. Via Venetian 21. 1.20133 Milano. ITALY

[@] **2003 by Organic Preparations and** Procedures **Inc**

HETEROSUBSTITUTED CARBOCYCLIC *OGAMINO* **ACIDS. A REVIEW**

M. L. Gelmi* and D. Pocar

Istituto di Chimica Organica, Facoltd di Fannacia Centro Interuniversitario di Ricerca sulle Reazioni Pericicliche e Sintesi di Sistemi Etero e Carbociclici Università di Milano, Via Venezian 21, I-20133 Milano, ITALY

INTRODUCTION

In recents years,' increased attention has been focused on the preparation of carbocyclic α -amino acids (Acc) characterized by α , α -substitution which improves stability towards hydrolytic metabolic degradation. Furthermore, the conformational rigidity and the presence of a further substituent on the ring allow for the formation of stereoisomeric derivatives whose interest is related to the attempts to clarify the conformational role of the substituent in bioreceptorial interactions.^{1c,2} These structural features make carbocyclic amino acids important biological targets both **as** single molecules and **as** components of bioactive peptides in place of natural amino acids. In these cases, important changes in peptide conformation were observed and consequent change of interaction with enzymatic active site produces a change of bioactivity. $³$ </sup>

As evinced by the increasing number of papers in the literature, a great deal of effort has been expended in the preparation of carbocyclic amino acids unsubstituted on the ring or substituted with alkyl or **aryl** groups.' Recently, increased attention **as** been focused on the preparation of the above compounds having one or more heterosubstituents linked to the ring. Heteroatoms such as oxygen, sulfur and nitrogen at the β - or γ -position allow the acquisition of constrained amino acids in which the skeleton of a natural amino acid is included. The presence of more **than** one heteroatom allows to change the polarity of the amino acid with consequent change of bioavailability. For example, a new and very promising research field is related to the preparation of hydroxy- and polyhydroxycarbocyclic amino acids, which **are** considered mimetic of carbohydrates but are characterized by a greater metabolic stability. The presence of a phosphonate group is very important in amino acids having agonist or antagonist activity against amino acid receptors in the **CNS** system. Furthermore, heterosubstituted carbocyclic amino acids have been used **as** the key starting materials for the preparation of other classes of non-heterosubstituted amino acids, which are usually bioactive molecules, exploiting the reactivity of the heterosubstituent.

I. GENERAL CONSIDERATIONS

The aim of this review is to present an as complete description as possible and classification of synthetic methodologies of the **known** carbocyclic amino acids (generally defined as Ac,c) bearing one or more substituents linked to the carbocyclic ring through a heteroatom. To facilitate the location of **an** item of interest, the carbocyclic amino acids are classified according to ring size. Accordingly, the main sections consider 1-aminocyclopropane- (Ac_c) , 1-aminocyclobutane- (Ac₄c), 1 -aminocyclopentane- (Ac₅c), 1 -aminocyclohexane-(Ac₆c), 1 -aminocyclohep $tane-1$ -carboxylic acids (Ac,c) , respectively. Mention is made also of the more representative classes of the corresponding bicyclic compounds, *i.e.* 2-aminobicyclo[3.1.0]hexane- (Ac_{13.1.0}c), 2-aminobicyclo[2.2.1]heptane- (Ac_[2.2.1]c) and 2-aminobicyclo[2.2.2]octane-2-carboxylic acids $(Ac_[2.2.2]c).$

A further classification is by the different heteroatom substituents. Thus in each chapter, sections are to be found dedicated to amino acids substituted with halo, nitrogen, oxygen, sulphur and phosphorus groups. For each compound or group of compounds, methods of synthesis are detailed.

Clearly, the above classification does not allow the grouping of the synthetic methodologies. However, this is of little importance because only a few general strategies were used to obtain the basic ring of 1 -aminocycloalkane- 1 -carboxylic acids. Many methodologies are specific and, though important by themselves, they have been used only in isolated cases. Accordingly they cannot be incorporated in general schemes. The more general and wide used synthetic processes **are** four and are indicated **as** methods A, B, C, D listed below. In the text, for each described procedure, the specific method has been indicated whenever possible.

1. Transformations and derivatization reactions of pre-existing cyclic compounds (Meth. A). This method was used mainly for the preparation of Ac₄c, Ac₅c, Ac₆c, Ac_[22,1]c and $Ac_[2,2,2]c$ compounds. The good availability of suitably functionalised starting compounds is essential.

2. *Cyclocondensations reactions of open-chin compounds (Meth. B).* Not surprisingly, this methodology is the most general one and offers a large number of possibilities to easily produce **all** the rings from the three-membered to the six-membered and also [3.1 .O]-membered ones. The necessary functional groups are easily created during the cyclization reaction.

3. *Cycloaddition reactions (Meth. C).* They appear to be used as the method of choice for the direct or indirect obtention, in stereocontrolled way, of three-membered rings (diazo compounds and alkenes), six-, [2.2.1]- and [2.2.2]-membered rings, (Diels-Alder reaction of substituted dienes with functionalised dienophyles).

4. Ring closing metathesis (RCM) (Meth. D). This procedure has been used for

preparing rings from five-membered to seven-membered.

In general, the functionalization with the amino acid moiety of the ring was done: a) both for methodology A and B using *i)* the Strecker or Bucherer-Berg reactions, when starting from a ketone or *ii)* by the classical Hofmann or Curtius rearrangements when starting from dicarboxylic acid derivatives;

b) for methodology C using *i)* conditions a) indicated above or *ii)* using a dienophile containing the amino acid moiety;

c) for methodology D using chiral bislactim compounds and hydrolysis after RCM reaction.

II. 1-AMINOCYCLOPROPANE-1-CARBOXYLIC ACIDS

Two general and useful synthetic approaches have been used for the prepamtion of **1** aminocyclopropanecarboxylic acids, namely the reaction of olefin derivatives with diazo compounds (Meth. C) and the intramolecular cyclocondensation reaction (Meth. B). As substituents, both halogen, oxygen and sulfur atoms have been linked to the ring. Cycloaddition reactions of diazomethane to readily available heterosubstituted alkylideneoxazolones *(Scheme* 3) is probably the method of choice for producing cyclopropane amino acids. Procedures are straightforward and yields are generally satisfactory. In general, a medium *cidrruns* diastereoselection is observed in favor of the *cis* isomer. Using Meth. **C** a single example of chiral synthesis of Ac,c compounds was reported starting from chiral acrylates (*Scheme 3*). Two examples on the use of Meth. B were reported. In this case starting from **a** chiral chain good results in term of enantioselection were achieved *(Scheme* 7). It is to be underlined that with the exception of halogen-substituted compounds, no examples of non protected amino acids were reported probably due to the unstability of heterosubstituted rings.

1. Halo Ac,c Compounds

The main synthetic approach to 2-halo compounds follows methodology **C** and reaction of a halocarbene with an olefin or reaction of a haloolefin and diazomethane were used. Monofluoro- and **2,2-difluorocyclopropane** carboxylic acids have been prepared by relatively simple procedures *(Schemes 1* and *2).* However, only in the second case **an** optically active compound was obtained. Recently⁴ the synthesis (Meth. C) of diasteromeric 2-fluoro-1-amino-cyclopropane- 1-carboxylic acids *5* (FAc,c) was reported starting from the 2-aryl-3-fluoroacrylate 1 which was reacted with diazomethane giving a mixture of trans- and cis-cyclopropane carboxylic esters $2 \text{ (trans/cis} = 3:1)$ (Scheme 1). After transformation of the ester group into the protected amino group (compounds 3), the aryl group was oxidized giving compounds **4** which were hydrolyzed to amino acids *5.*

Downloaded At: 19:57 26 January 2011 Downloaded At: 19:57 26 January 2011

a) CH2N2/Et20,0°C; b) **hv,** acetone (90%); c) NH2NH2, H20, EtOH; d) HCI, NaN02, then ROH, reflux (54%); **c)** $\text{NaIO}_4/\text{RuCl}_3$, $\text{CCL}_4/\text{MeCN/H}_2\text{O}$ or $\text{O}_3/\text{H}_2\text{O}_2$, CH_2Cl_2 , reflux (95%); **f)** $R = t - Bu$: **aq.** HCVAcOEt; R = Et: HBr/AcOH, reflux (30%) **Scheme 1**

The enantioselective synthesis of **R-l-amino-2,2-difluorocyclopropane-l-carboxylic** acid **115** was achieved according to Meth. C starting from diacetate **6** which was reacted with difluorocarbene giving *7 (Scheme* 2). The enantioselective synthesis of R-1-amino-2,2-difluor was achieved according to Meth. C starting from diacet
carbene giving 7 (Scheme 2).
 $\begin{array}{ccc}\n & a,b \\
\hline\n0^{\text{OAc}} & \xrightarrow{a,b} \\
\hline\n0^{\text{OAc}} & \xrightarrow{c} \\
\hline\n\end{array}$

a) ClF₂CCO₂Na, diglyme, 180°C; b) K₂CO₃, MeOH/H₂O (45%); c) Vinyl acetate, PS (96%); d) Jones oxid.; e) DPPNt-BuOWEA, benzene, reflux; **f)** aq. HCYAcOEt (46%)

Scheme 2

The lipase-catalyzed asymmetric acetylation with Amano **PS** from *Pseudomonus cepuciu* of the pro-chiral alcohol **7** gave the chiral monoacetate 8 in excellent yield and enantioselectivity (88% ee). Compound 8 was transformed into **11** according to the synthetic procedure depicted in Scheme 2.

The synthesis of chloro and bromo spiro compounds **13a,b** *(Scheme 3),* the precursors of 2-chloro- or 2-bromo Ac,c compounds, was reported⁶ using the 4-halomethylene-2-phenyl-5(4H)oxazolones **12a,b.** The reaction resulted in the formation of a mixture of *cis-* and *truns*spiro compounds **13a,b.** In both cases the diazomethane insertion by-products **14a,b** were obtained. The formation of compounds **13a** was favored starting from the chloro derivative **12a** (73%; *cis/trans* = 20:53). This ratio has been revised *(cis/trans* = 53:20).^{6d} By way of contrast, starting from **12b,** compounds **13b** were obtained as the minor products (17%; *&/trans* = 152.5). Two other synthetic approaches to bromo derivatives were reported. The 2-bromo Ac,c derivative **22'** is the synthetic intermediate for the preparation of 1 **-amino-2-methylenecyclo**propane- 1 -carboxylic acid **23,** an Ac,c deaminase inhibitor. As depicted in *Scheme 4* compound

a) CH_2N_2 , CH_2Cl_2 ; b) EtOH, DMAP (60-77%); c) I₂, EtOH (97%); d) RBr, CTACl, NaOH, THF/H₂O, amino-iminomethanesulfinic acid (67-74%); e) (-)-menthol, (Et₂SnCl)₂O, benzene, reflux (78%); f) 150°C (65%)

Scheme 3

21 was obtained (Meth. C) from 2-bromoprop-2-en-1-ol which was reacted with diazomalonate. The amino acid derivative 22 was obtained from compound 21 by the Curtius reaction.

A different approach (Meth. B) to introduce the bromine atom on the cyclopropyl ring was followed by Wick et al. (Scheme 5).⁸ By reaction of epichlorohydrin and dimethyl malonate the lactone 24 was obtained which was selectively transformed into the monoamide 25 after

a) MeONa, MeOH; h) NH₃/MeOH; c) TBDMSCl; d) Pb(OAc)₄, *t*-BuOH, then Bu₄NF; e) RuCl₃/NaIO₄ *(38%);* f) NMM/ClCO₂*i*-Bu, -15°C; g) NHPT, -15°C; h) ALBN/BrCCl₃, reflux (52%); i) 6N HCl, 80°C *(cis:* 68%; trans: 75%) **Scheme ⁵**

reaction with ammonia and protection of the hydroxy group. The amido group was transformed into the protected amino group and, after deprotection of the hydroxy group and its oxidation under mild conditions, compound **26** was obtained. By the radical decarboxylative bromination, according to Hunsdiecker reaction, the bromo compounds *cis-* and **trans-27** (3: **1)** were obtained and separated. Complete hydrolysis allowed the isolation of the amino acids *cis-* and *trans-28*. Comparing this synthetic procedure to those reported in Scheme 3 it has to be observed that the overall yields of protected amino acid derivatives **13b** and **27** are comparable (less than **20%),** but the diastereoselection is better when starting from oxazolone.

The formation of dichloro amino acids **30a,b,** as by products, was reported by Schwas *et* **~1.~** from methyl acetylenedicarboxylate and dimethylallylamines **29** in CHCI, as the solvent and a dichlorocarbene donor (Meth. C) (Scheme 6).

2. *Oxygen Ac,c Compounds*

No free OH-substituted amino acids have been reported. Following Meth. B, Cativiela et al.¹⁰ reported the diastereoselective synthesis of chiral (1S,2R)-2-benzyloxy derivative 33 (Scheme 7) from the corresponding nitrile **32a.** The latter was obtained with compound **32b** in good diastereomeric excess by an intramolecular cyclization of chiral compounds **31a** or **31b (32a/32b, 98:2)** under basic conditions. However, nor nitriles **32** nor the amide **33** have been hydrolyzed to the corresponding acids.

A similar situation is depicted in Scheme 3. According to Meth. C, the 4-acyloxymethylene-2-phenyl-5(4H)oxazolones 12c,d^{6c} can be used for the preparation of 2-acyloxy-1-aminocyclopropanecarboxylic acid derivatives **13c** and **13d** which were obtained in 76% and 73% yield, respectively which low diastereoselection *(cidtruns* = 3442 and 43:30, respectively).

3. *Sulfur Ac,c Compounds*

Cyclopropane amino acids bearing a **RS** substituent have been prepared in fair number by relatively short procedures and with appreciable diastereo- and enantioselectivity. Examples are reported in Scheme *3."* In fact, compounds **13,15,17** were obtained according to Meth. C from 4-alkyl- or 4-phenylsulfanylmethylene-2-phenyl-5(4H)-oxazolones $Z/E-12e-h$ which, on reacting with diazomethane (affording spirooxazolones 13e-h: *cis/trans* = 3:1 to 5:1 ratio) and ethanolysis, were transformed into compounds **15e-h.** The trityl derivative **15h** is useful for the preparation of a large number of 2-alkylsulfanyl derivatives **17.** In fact, on treating **cis-15h** with iodine, the disulfide derivatives **16** were obtained and alkylated under FTC conditions in the presence of **amino-iminomethanesulfinic** acid, **as** the reductant, giving the acids **17a,b.** The same authors¹² have achieved the asymmetric synthesis of the 1-amino-2-S-trityl Ac₃c derivatives from chiral (-)-menthy1 aminoacrylate **(Z)-18a** obtained **as** depicted in Scheme 3. The reaction of compound **18a** with diazomethane gave two stable diastereomeric pyrazolines **19a** and **19b** (1:1.4) with *cis* relationship between nitrogen and sulfur atoms. On melting at 150" for ten minutes, the pyrazolines **19a** and **19b** gave cyclopropyl compounds **1R,2R-2Oa** and **1S,2S-2Ob.** This transformation occurs with retention of configuration at *C-a* of the amino acid. The synthesis of enantiomeric compounds of **2Oa** and **2Ob** was effected.

III. 1-AMINOCYCLOBUTANE-1-CARBOXYLIC ACIDS

The 3-heterosubstituted Ac_ac compounds are highly representative and the most common method for their preparation consists in the use of functionalized heterosubstituted cyclobutanes *(i.e.* heterosubstituted cyclobutanones or cyclobutanecarboxylic acids from **a** cyclocondensation reaction of heterofunctionalized chains; Meth. B). A different approach takes advantage of the use of heterofunctionalized bicyclo^[1.1.0]butanes (Meth. A).

1. Halo Ac,c Compounds

The method for preparing 3-flUOrO derivatives **was** applied only to labeled halogen derivatives and this approach (triflate exchange) (Scheme 8) can be used also for larger-ring compounds, thus appearing of general scope. Diastereomeric separation is possible. Similarly, a

a) NaH, CH₂(CO₂Et)₂, dioxane; b) NH₄OH; c) NaOCl, H₂O, 0°C (60%); d) KOH/EtOH, H₂O, then 6M HCl (72%); e) $Ba(OH)_2$, then H_3O^+ ; f) (t-BuCO)₂O, MeOH, TEA, 60°C; g) R = Me: CH_2N_2 , Et₂O (n = 1: 30%); h) R = Et: DPPA, TEA, toluene 80°C, then *t*-BuOH, 110°C (58%); i) H₂, Pd/C, MeOH (89%); 1) (Tf)₂O, CH₂Cl₂, Py (64%); m) $X = {^{18}F: K^+}/{^{18}F}$; $X = {^{123}I: [{^{123}I}]}$ -Nal; **K222, H₂O/MeCN;** n) 6M HCl, $X = F$: 30%; o) $R = Me$: PCC, DMF; p) $R = Et$: $RuO₄$, NaI $O₄$, H₂O/CCl₄ (80%); **q**) NH₂NH₂, DBN, EtOH, reflux; r) 3% H₂O₂, [¹²³I]-NaI, then 0.1N HCl; **s)** BnONH2, NaBH3CN. MeOH (50%) **Scheme**

3-iodo compound was prepared. Another way **to** the same iodo compound was indicated, probably of low preparative value (no indication of yield and absolute configuration). The new tumor-avid labeled **1 -amino-3-fluorocyclobutane-** 1 -carboxylic acid (FAc4c) *truns-39a* has been synthesized13 from **34a,** obtained by reaction of diethyl malonate and 1 -chloro-2-benzyloxy-3bromopropane (Meth. B). Compound **34a** gave a mixture of diastereomeric hydantoins **35a** (70%, **cis/trans** = **7525)** which were separated. The main compound **cis-35a** was transformend into the aminoester **cis-37a** *(Scheme* 8). Deprotection of the oxygen atom and its functionalization with a triflate group gave **cis-38a.** Reaction with fluoride ion and hydrolysis gave the labelled compound **trans-39a.** In line with the above synthetic **procedure** the labelled **trans 1** amino-3-iodocyclobutane-1-carboxylic acid 39d was prepared.¹⁴

The same authors¹⁴ report the synthesis of the labelled unsaturated iodo derivative 42a from ketone **Ma,** through the corresponding hydrazone **41a.** Treatment of **this** compound with iodine and hydrogen peroxide, deprotection of amino and carboxylic groups gave **42a**.

2. Nitrogen Ac₄c Compounds

A mixture of **cis** and **trans** protected **1,3-diaminocyclobutane-1carboxylic** acid derivatives **44** (1:1) were prepared from ketone **40d** by reacting with *O*-benzylhydroxylamine in the presence of cyanoborohydride. They *are* precursors of the new selective methatropic glutamic acid (mGluRSa) receptor agonists *45. (Scheme* **8)15** Compound **4Oa** was used for the preparation of hydrazone **41a** *(Scheme* **8).14**

3. Oxygen Ac4c Compounds

An alternative way to 3-oxygen substituted protected amino acid 37 (37d $R = Et$) was reported from **34a** *(Scheme* **8)16** which consists in selective hydrolysis to the monocarboxylic acids **cis-** and **trans-36.** The Curtius reaction on this **mixture** followed by reaction with **r-BuOH** produced the corresponding N-protected diastereomeric amino acids **37d.** The preparation of compound **37d via Curtius** transposition is more efficient (43% overall yield) to respect of the preparation of **37a** (1 3%) **via** hydantoin **35.** The mixture of compounds **37d,** after deprotection of the oxygen atom and oxidation afforded the 3-keto derivative **4Od as** precursor for the preparation of bioactive amino acids **cis-** and **trans-43.**

A synthesis of uncommon dihydroxy-Ac₄c compounds was reported¹⁷ from oxazole 46 dimerized to **47** *(Scheme* 9). The treatment of the latter with hydrogen bromide gave the protected amino acids **48** (quantitative yield, **cis** relationship between the nitrogen and oxygen atoms). Though interesting in itself, the synthetic procedure to **48** is not of general scope.

4. Sulfur Ac,c Compounds

Several derivatives of **3-phenylsulfonyl-1-aminocyclobutane-1-carboxylic** acid, which are the precursors for the preparation of the bioactive 1,3-dicarboxylic amino acids **60** and **61,** were prepared by rearrangement of **3-(phenylsulfonyl)bicyclo[** 1.1 .O]butane- **1** -carboxylic acid derivatives **49** (Meth. A)¹⁸ (Scheme 10).

 $R_1 = H$, Me, *i*-Pr, $R_2 = OH$, OMe, OEt, NHBn, N(CH₂)₅. **50,51** $R_3 = NHBn$; **52,53** $R_3 = N_3$

a) BnNH₂, 140°C (56-90%) or NaN₃, DMF or NMP, 85°C, TMGA (80-95%); b) Na/Hg, MeOH/THF; *c*) **BuLi, -78°C, MeI, then method b; d)** $R_4 = H$ **: Pd/C, HCO₂NH₄, MeOH, 50°C (60-70%); e)** $R_4 =$ **COPh: method d, then PhCOCl(60-70%); f) BuLi, -78"C, COz, THF**

Scheme 10

The addition of a nitrogen nucleophile *(ie.* benzylamine or sodium azide) to **49** gave the diastereomeric amino compounds **50** and **51** or the corresponding 1-azido derivatives **52** and **53.** In many cases the addition reaction of amine was diastereoselective, the *trans-S,N* products *50* being preferred according to *cis* addition of the amino group and the proton delivered by the amino group itself. Instead, the addition of the azido group gave a mixture of *trans-S,N* and *cis-S,N* stereomers. However, when the azidation reaction was applied to the free acids **(49:** R_2 = OH) in DMF in the presence of tetramethylguanidine, only the *cis-S_nN* compound 53 ($R_2 = OH$, $R_3 = N_3$) was detected. As an example, the benzylamine derivative 51 ($R_1 = Me$, $R_2 = NHBn$, R_3)

= NHBn) was desulfurated producing *54* or alkylated at C-3 and then desulfurated producing **55.** The best method to reduce the azido group in compounds *52* and *53* consisted of the use of ammonium formate in methanol in the presence of palladium giving the amino acid derivatives **56** and **57** $(R_4 = H)$. Finally, C-3 carbon on the cyclobutyl ring was carboxylated by treating both the azido acids 52 or 53 and the N-protected compounds 56 or 57 $(R_4 = \text{COPh})$ with n-BuLi at -78° and carbon dioxide. A mixture of *cis*- and *trans*-acids **58,59** ($R_3 = N_3$, NHCOPh) was obtained. They were transformed into the dicarboxylic acids **60** andor *61.*

5. Phosphorus Ac,c Compounds

The reported procedure, useful for obtaining phosphono amino acids 64 (Scheme 11),¹⁹ which are known to act **as** antagonists at excitatory amino acid receptors, is relatively simple, but

a) MePO(OEt)₂, BuLi, THF, -78°C; b) H₂Pd/C, then RuCl₃/NaIO₄, CH₂Cl₂ (45%); *c***) PMBNH**₂, NaCN, MeOH/AcOH (53%); d) CAN, MeCN, H₂O or H₂, Pd/C; e) 6M HCl (74-76%)

Scheme *11*

diastereoselection is moderate. In any case, separation of *cis* and *trans* isomers could be accomplished easily. They were synthesized (Meth. **A)** starting from **diethyl3-oxocyclobutylphospho**nate **62,** obtained from **l-chloro-2-benzyloxy-3-bromopropane.** The reaction of *62* with p-methoxybenzylamine and sodium cyanide producing a mixture of the diastereomers *cis-* and *trans-63 (cis/rrans,* 3:2). The isomers were separated and the p-methoxybenzyl group was removed under mild oxidative conditions using ceric ammonium nitrate or catalytic hydrogenation giving N-unprotected compounds which were hydrolyzed to amino acids *cis-* and *tram-64.*

IV. 1-AMINOCYCLOPENTANE-1-CARBOXYLIC ACIDS

A large number of heterosubstituted Ac,c compounds were synthesized with functionalization both at C-2 and C-3. Furthermore poly-heternsubstituted compounds were reported. The most common starting materials for 2- or 3-heterosubstituted $Ac_{c}c$ compounds are cyclopentanones or cyclopentanecarboxylic acids (Meth. **A)** which were functionalized with a heteroatom. Both RCM reaction (Meth. D) and intramolecular cyclocondensation starting from heterofunctionalized chains (Meth. B) were reported. In this case, as a key precursor, a chiral sugar was used for the preparation of polyhydroxylated compounds.

1. Halo Ac,c Compounds

2-Halo- 1 **-aminocyclopentanecarboxylic** acids are scarcely considered: a single synthesis (besides the 1-chloro isomer) of 2-chlorocyclopentane derivative *8920 (Scheme 15, see* p. 156) was reported (low yields, 16%) from aziridine **88**, obtained from ethyl α , β -cyclopentenecarboxylate *87* and **3-acetoxyaminoquinazolinone.** The method for cyclobutane analogs (Scheme 8) was applied to 5-membered labelled 3-fluoro derivative *trans*-39b and iodo derivative $42b$ (n = 2).¹⁴ In this case too, yields and configuration are unknown.

2. Nitrogen Ac,c Compounds

A relatively straightforward synthesis (no yield) for **1-amino-2-nitrocyclopentanecar**boxylic acid *67,* a metabolite of *Aspergillus wentii,* a plant growth inhibitor, was reported by a fermentation process using the above microorganism.²¹ The synthesis was also done starting from cyclopentanecarboxylic acid 65^{22} which was transformed (Meth. A) with iodine and N₂O₄ into intermediate **66** and then into amino acid *67 (Scheme 12).* As shown in *Scheme* 8 the hydrazone derivative $41b$ $(n = 2)$ was obtained. The yield was not given.

Scheme *12*

3. Oxygen Ac,c Compounds

Hydroxy substituted 1 -aminocyclopentane- 1 -carboxylic acids are the most studied family. Syntheses of both 2-, 3- and polyhydroxy derivatives are reported.

i. Substitution at C-2

Different synthetic approaches to 2-hydroxy Ac_sc compounds were described. The achiral stereomeric 2-OH compounds *73* were prepared according to *Scheme* 13 both starting from **1 -cyclopentene-1-carboxylic** acid *68* and ketone *71.* Better results were observed starting from **71** both in term of yield and diasteroselection. When *68* (Meth. A) was reacted with Hg(OAc), in methanol and then with bromine, compound *69* was obtained. The reaction with ammonia (compound *70)* and hydrolysis gave the amino acids *cis-* and *trans-73* in **45%** overall yield and in about 1:1 ratio.²³ Instead, the use of the Bucherer synthesis (Scheme 13) led to a mixture of diastereomeric amino acids *cis-* and *trans-73* with improved yields (60%) and diastereoselection (84:16, the major product having the carboxy and hydroxy groups *cis*).²⁴ Ketone 71 (Meth. A) on reaction with KCN and (NH_4) , CO₃ afforded the hydantoin 72 which was hydrolyzed under basic conditions and demethylated to *73.*

a) Hg(OAc)₂, MeOH; b) KBr, Br₂, hv, H₂O (47%); c) NH₄OH; d) 49% HI, reflux; e) KCN, $(NH_4)_2CO_3$, EtOH (69%); f) Ba(OH)₂, H₂O; g) 48% HBr

Scheme 13

Two chiral approaches to 2-hydroxy amino acids, having N,O-cis relationship, characterized by good diastereo- and enantioselection (Schemes 14 and 38) have been reported. Chiral compound 1R,2S-73a (41% overall yield from 78) (Scheme 14) was reported by Ohfune et al.²⁵ starting from 1,1-dimethoxy-2-cyclopentanol (77) (Meth. A). Reaction of 77 with N-t-butoxycarbonyl-L-phenylalanine gave 78 and deprotection of both ketone and amino functional groups

a) Boc-L-Phe, WSCD, DMAP (90%); b) p-TSA, acetone; c) TFA (89%); d) $n = 1$: MgSO₄, AcONa, MeCN, then TMSCN, ZnCl₂, i-Pr₂O (78%); n = 2: NaCN, 2-propanol; e) t-buthyl hypochlorite, THF, 0°C, then TEA; f) 12M HCl (60%); g) Jones reagent, acetone Scheme 14

produced compound **79** which was condensed to intermediates 80. The addition reaction of cyanide ion resulted in good diastereoselection $(5S, 6S/5S, 6R = 92:2)$ and the major isomer 81 was isolated. Removal of the chiral phenylalanyl moiety from the major diastereomer and hydrolysis gave enantiopure amino acid **73a.** The synthesis of the lS.2R-enantiomer of **73a** was also reported using D-phenylalanine **as** chiral auxiliary. The synthesis of chiral P-hydroxy unsaturated Ac,c compound **232** (Scheme 38, see p. 175) was reported starting from chiral bislactim **220a** (Meth. **D).26** Hydrolysis gave amide **231a** which was protected at nitrogen atom giving **231b.** Using the ring-closing methatesis reaction, protected compound **232** was obtained **(40%** overall yield from **220a).**

The 2-oxo-Ac_sc compound 93a²⁰ was obtained using 3-acetoxyaminoquinazolinone (QNHOAc) as nitrogen donor. By reacting with ethyl **2-0x0-cyclopentanecaboxylate 91a** (Scheme 15) compound **93a** (77%) was formed through the 2-hydroxyaziridine intermediate **92a.**

ii. Substitution at C-3

All stereomeric 3-hydroxy compounds and their derivatives have been prepared, with some attention to stereoselective approaches. A simple and highly diastereo- and enantioselective method exists for the preparation of the **S,S** isomer (Scheme *16)* and a satisfactory procedure has been devised to produce both the **S,S** (the latter less efficiently) and the *S,R* isomer (Scheme *17)* from relatively common intermediates. According to Scheme 8 (Meth. B) a mixture of achiral cisand trans-3-hydroxy compounds (0-benzylated) **37b** were obtained. The **cis-37b** derivative was transformed into the triflate derivative *cis*-38b and from the *cis/trans* mixture the corresponding ketone 40b was isolated. Yields were not reported.¹⁴ The 1S,3S-1-amino-3-hydroxycyclopentanecarboxylic acid derivative 97a (62% from 94) was reported by Ma et al.²⁷ (Scheme 16) starting from chiral compound **94,** which was cyclized (Meth. **B)** to the cyclopentyl derivative **95** by reaction with diethyl malonate. The ethoxycarbonyl group trans to the DPM group was selectively hydrolyzed.

a) $CH₂(CO₂Et)₂$, EtONa/EtOH (70%); b) NaOH, EtOH; c) DPPA, TEA, benzene, reflux, then EtOH (88%); d) H₂/Pd, EtOH (100%)

Scheme 16

The corresponding acid was converted into the enantiopure carbamate $1*S*,*3S*$ -96 by means of a Curtius rearrangement (only 3% of the *cis* isomer **was** formed). The **DPM** group is essential for enantioselectivity during the ester hydrolysis. In fact, when benzyl was used as protecting group only a 1:3 *cis/trans* stereomeric ratio was obtained. The hydroxy protecting group was removed giving *N,O-trans-97a* which is the precursor of the bioactive 1-amino-1,3cyclopentanedicarboxylic acids 98a,b. A different synthetic approach to chiral 3-hydroxy-Ac_cc derivatives was followed by Hodgson²⁸ (Scheme 17) starting from the 1-amino-3-cyclopentenecarboxylic acid derivative 99 (Meth. A) taking advantage of the selectivity in the hydroboration reaction of the double bond assisted by the amide moiety.

a) (COCl)₂, CH₂Cl₂; b) NaN₃, acetone/H₂O, then *t*-BuOH, SnCl₄, toluene (71%), c) BH₃, THF, then H_2O_2 , NaOH; d) BF_3E_2O , [(+)-IpcBH₂]₂TMEDA, THF, -40°C, then H_2O_2 , NaOH (74%); e) AcOH, Ph₃P, THF, DEAD, then K₂CO₃ (58%); f) MCPBA, TEA, CH₂Cl₂ (59%); g) *s*-BuLi, (-)sparteine, Et20, -78°C (18%); **h)** H2/Pd, EtOH (78%)

Scheme 17

GELMI AND POCAR

This route appears to be less diastereoselective, but allowed to obtain the **R,S** isomer. Compound **lOOa** was prepared from 99, *via* the acid chloride and subsequent Curtius rearrangement. When reacted with BH, and then oxidized, compound **lOOa** gave, after a few minutes, a mixture of diastereomeric hydroxy derivatives **lOla** and **97b** in 77:23 ratio. Prolonging the reaction time favored the formation of compound **lOla (101d97b:** 30 min: 84:16; 17 h: *955)* with lowering of yield (10 min: **84%;** 30 min: 73%). Different reagents and reaction conditions were tested for the asymmetric hydroboration reaction of **1OOa.** The best reagent was (+)-IpcBH, which gave compound **lS,3R-lOla** (48% ee; **94%)** having the *cis* relationship between nitrogen and oxygen atoms. The enantiomer $1R$, 3S-101a could be obtained using $(-)$ -IpcBH₁. Alcohol **1S,3R-lOla** was converted into chiral compound **1S,3S-97b** (56% from **1OOa)** by Mitzunobu isomerization at C-3 carbon. An alternative, but less efficient diastereo- and enantioselective synthesis of compound $(-)$ -101a (Meth. A) was followed by the same authors starting from 100a which was epoxidized giving a mixture of two diastereomeric epoxides **102a** and **102b** (85:15). The major isomer **102a** was rearranged to the allylic alcohol 1S,3R-103 in very low yield (18%) and enantiomeric excess (33%) using s-BuLi/(-)-sparteine. After hydrogenation, the alcohol **103** was transformed into chiral compound *IR,3S***-101a**, the precursor of dicarboxylic amino acid **98a.** The synthesis of 3-keto amino acid **123** was reported by oxidation of compounds **97** or **lOlb** *(Scheme* **22,** seep. 161).'s

iii. Substitution at C-3,4

Methods to 3,4-dihydroxy compounds **are** reported, but only the compounds having *cis* relationship between the OH substituents are known. Moreover, in two cases *(Scheme* **18)** only protected OAc and OMe derivatives have been described. A single example of free OH substituents has been reported *(Scheme 19),* but without indication of yields and diastereomeric ratio. **An** original diastereoselective synthesis of 1 **-amino-3,4-dihydroxycyclopentane-carboxylic** acids **108** *(Scheme f8),* in which the two hydroxy groups are *cis* with respect to the amino one,

a) 2% Ru(II), DEC; **b**) OsO_4 , NMO, Me₂CO, H₂O, 0°C (71%); **c**) NaH, MeI, DMF/THF; d) 0.1 M $TFA, MeCN/H₂O; e) Ac₂O, DMAP, CH₂Cl₂$

Scheme 18

was reported by Hamme?9 starting from the chiral spiro compound **105a** obtained from *alky*lated bislactim **104a** through Ru(II)-catalyzed ring-closing metathesis reaction (RCM) (Meth. D). The addition of catalytic osmium tetroxide in the presence of N-methylmorpholine N-oxide afforded the diasteromeric dihydroxylated compouds **106a** and **1Wa** in **3:l** ratio. Compound **106a** was transformed into the corresponding dimethoxyamino acid **lO8a (35%) through** methylation and hydrolysis. The diacetoxyamino acid **loSb (60%)** was **also** obtained by hydrolysis of **106a** followed by acetylation.

A different synthetic approach30 to 3,4-dihydroxy Ac,c compounds **108** and **110** (Scheme *19),* the inhibitors of zinc metalloendopeptidases, was reported from cyclopentene derivatives **100** (Meth. A) by reaction with osmium tetroxide. The reaction formed a mixture of diastereomeric compounds **108** and **110.** Yield and diastereomeric ratio were not reported.

iv. Polysubstitution

Several polyhydroxy compounds (diastereomeric **l-amino-2,3,4,5-tetrahydroxycyclo**pentanecarboxylic acids (Scheme *20)* have been prepared, starting from sugar-like precursors. **Procedures** are relatively straightforward in view *of* the complexity of the products. Compounds **11S3'** were obtained from chiral azidolactones **112a-c.** Lactones **112a** and **112b** were obtained from epimeric aldehydes **111** by an intramolecular closure of the azido anion (Meth. B) (Scheme *20)* in a different ratio depending on the base used to generate the anion $(F: 112a/112b = 58:15; N;$ **112d112b** = **22:66).** The epimerization of lactone **112b** to lactone **112c** occurred in ammonia through an aldol equilibration. Lactone **112a** was reacted with aqueous potassium carbonate giving the hydrolysis product **113a** which was transformed into the amino acidic compound **115a** (98% overall yield) after azido group reduction (intermediate **114)** and removal of the protecting group. The same lactone **112a** was transformed into diastereomeric amino acid **115b** using the same reactions, but with a different sequence: deprotection of hydroxy groups (compound **116a),** isomerization of the hydroxy group (compound **116b),** reduction of the azido group and hydrolysis. *Amino* acid **115b** was obtained **as** the major isomer together with a minor amount of **115a.** Starting from either **112b** or **112c** *via* the same reaction sequence (deprotection of hydroxy groups, reduction of azido function, hydrolysis of lactone pup), a mixture of amino acids **115d115d (70%, 1:8)** and **115e/115f** (54947, **6:l)** was obtained, respectively. In contrast, starting from 112c by hydrolysis of the lactone group as the first step, followed by reduction and deprotection reactions, the amino acid **115f** was the only diastereomeric compound (79% overall yield). The same authors reported on the synthesis of amide or hydantoin derivatives of amino acids **115a,d,f.32**

a) NaF, MeCN, 18-crown-6, -6°C **(73%),** or NaN3, MeCN, 18-crown-6 (88%); b) liq. **NH,** (82%); c) K2C03, H20; d) H2/Pd, **H20;** e) CF~COZH, H20; f) AcONa, DMF **(94%); g)** TEA/H20 **(95%)**

4. *Sulfizr Ac,c Compounds*

The 2- and **3-alkylthiocyclopentanones 117a** and **120a,** prepared as described in Scheme *21,* were used for the preparation (Meth. **A)** of diastereomeric cysteino- or methioninolike amino acids 119a³³ and 122a.³⁴ The Bucherer-Berg and Strecker reactions were respectively used for the preparation of intermediates **118a** and **121a** which were hydrolyzed to the corresponding amino acids **119a** and **122a.** The diastereomeric amino acids **122a** were obtained in **¹**: **1.3** ratio and were separated. The stereochemistry was not assigned.

5. Phosphorus Ac,c Compounds

Both unsaturated *(Scheme* 22) and saturated *(Scheme* 23) 1-amino-3-phosphonocyclopentene- and cyclopentane- 1 -carboxylic acid were reported. Yields are satisfactory, but diastereoselection, when applicable, is absent. No enantioselective methods have been reported till now. Unsaturated componds³⁵ **128** and **129** (Scheme 22) were obtained from the mixture of

TEA, DMF, (Ph3P)4Pd, 70°C, then flash Chromatography **(126:** 80%; **127: 12%);** e) 6N HCI **(128:** 97%; **129:** 57%) **Scheme 22**

stereomeric methyl esters **97,101b** (Meth. A). The corresponding ketone 123 was obtained by oxidation and converted into a regioisomeric mixture of enol triflates **124** and **125.** The Pdcatalyzed addition of diethylphosphate afforded in very good yield **the** corresponding compounds **126** and **127** (6.51) which were separated and hydrolyzed to amino acids **128** and **129. A** different approach was followed for the synthesis of the corresponding saturated diasteromeric compounds **134a** and **134b** *(Scheme* **23).36** Diastereomeric compounds **131a** and **131b (80%,** 1:1) were obtained using the Strecker reaction from ketone **130a** (Meth. A) obtained from cyclopentenone through a hydrophosphinylation reaction **(73%).** Their separation was achieved by acetylation and chromatography **(133a: 46%** and **133b:** 24%). The hydrolysis gave the corre-

sponding amino acids **134a (95%)** and **134b (80%).** The 3-phosphino Ac,c compound **278** was prepared by Johnson *et* al. according to Scheme *46* but the reactions and yields were not detailed.³⁷ **EXECUTE:** CO₂H
Solution Scheme 10 and 134b (80%). The 3-phosphino Ac₅c compound 278 w
 ACHN₂ **CO₂H ACHN**₂ **CO₂H ACHN**₂ **CO**₂H **ACHN**₂ **CO**₂H **CO**₂H **CO**₂H **CO**₂H **CO**₂H **CO**₂H **CO**₂H **C**

a) NH₄Cl, NH₄OH, NaCN; b) Ac₂O, Py, then flash chromatography; c) 6N HCl **Scheme 23**

V. 1-AMINOCYCLOHEXANE-1-CARBOXYLIC ACIDS

Several syntheses of Ac,c compounds functionalized at C-2, C-3 and **C-4** or polyfunctionalized with heteroatoms are known, the derivatives substituted with an oxygen atom being the more representative. The two main synthetic procedures adopted for their preparation made use of *i)* cyclohexanones bearing **an** heteroatom on the ring and using Strecker or Bucherer-Berg reactions for the introduction of the amino acidic moiety (Meth. A) and *ii)* of the Diels-Alder reaction (Meth. C). As for Ac,c compounds both oxazolones and aminoaaylates **are** the most common starting materials in the Diels-Alder cycloaddition reaction and the heteroatom can be linked to the diene or the dienophile.

1. Halo Ac,c Compounds

Few examples of halo substituted Ac,c compounds are reported in the literature *i.e.* the protected 2-chlorocyclohexane derivatives, which were transformed into the more interesting hydroxy derivative 143a, and the 4-iodo compound as intermediate of a synthetic pathway.

The two diastereomeric **l-amino-6-chlorocyclohex-3-ene-** 1-carboxylic acid derivatives **137a** and **138a** were prepared³⁸ according to methodology C (Scheme 24). EtAlC1, catalyzed the Diels-Alder reaction between chloromethyleneoxazolone **2-12a** and 2,3-dimethylbutadiene **(13fia)** giving a single diastereomeric adduct **137a** with *cis* relationship between the nitrogen and

propylene oxide **(50%)** Scheme 24

chloro atoms. It was transformed into ester **139.** Similarly, cycloadduct **137b** was obtained with good regioselectivity from isoprene **(136b).** It was possible to obtain the diasteromeric adduct **138a** from oxazolone **Z-12a** which was partially isomerized to **E-12a** (1: **1)** by W light. The reaction mixture was directly used for the cycloaddition reaction with **136a** and a mixture of cycloadducts **137a** and **138a** (6796, 54:46) was isolated. Treatment of adducts **137a** and **138a** with dimethylamine produced the corresponding amides **140** and **141** which were separated. As for Ac_4c and Ac_5c ring system, the labelled fluoro- Ac_6c compound *trans*-39c and the iodo compound **42c** (n = **3)** were reported. For their synthesis *see Scheme* **8.14** Yields and stereochemistry were not reported.

The chiral 3-acetoxy-4-iodo derivative³⁹ 144 (Scheme 25), an important intermediate for the preparation of constrained 4-hydroxyproline **145,** was obtained (Meth. A) from the corresponding oxazine **25Ob** *(see Scheme 40,* p. 177).

2. Nitrogen Ac₆c Compounds

Poor information exists about 2- and 3-amino compounds which are reported but without synthetically useful **data.** Instead, both diastereomers of 1,4-diamino acids (constrained omitine analogs) have been successfully prepared by traditional procedures *(Schemes 26-28).* The 2-amino Ac_6c compound 146 (Figure 1) was cited⁴⁰ and for its synthesis the Fondaker's procedure is indicated. However, no data were given for starting materials, intermediates and yields. The 3-amino $Ac_{\xi}c$ compound 147 was also reported⁴¹ for its anticonvulsant activity, but the synthetic scheme was not reported *(Figure* I).

Different synthetic approaches were used for the preparation of the isomeric 1,4 diamino acids **151** and **152,** which are ornitine analogues (Meth. A). The first synthesis was reported in 197342 starting from **4-toluenesulfonylaminocyclohexanone 148** *(Scheme 26).* Nitrile **149** was prepared in quantitative yield by Strecker reaction and hydrolyzed to the diasteromeric

a) KCN, NH₄Cl, MeOH/H₂O (100%); b) KCN, (NH_4) ₂CO₃ (90%); c) AcOH, 12N HCl, 120°C; d) MeOH, HCl Scheme 26

compounds **151** and **152 (60:40)** in very low yields. The use of Bucherer-Libe reaction followed by hydrolysis of hydantoin **150** gave compounds **151** and **152** in **60%** yield with inverted ratio **(20:80).** The *N,N-trans* configuration of isomer **151** was confirmed by spontaneous cyclization of its methyl ester hydrochloride to the bicyclic compound **153.** The above amino acids were also prepared from **4-acetamidocyclohexanone** via similar routes but yields were not given. A different approach to 4-amino Ac₆c derivatives *(Scheme27)* was reported⁴³ from 1,4-cyclohexanedione monoketal **154** (Meth. A) which was transformed by a reductive amination reaction into a series of amines **155.** After deprotection of the ketone group and protection of the amino function (intermediate 156a, $R_1 = Boc$), the amino acidic compounds 157a were obtained by the classical Bucherer-Berg procedure followed by hydrolysis. However, yield and configurations of the products were not reported. A diastereoselective route to compound **158** was used by Himmelsbach *et al.* (Meth. A) *(Scheme 27).** From **154** and benzylamine, compound **155 (R** = Bn, 72%)

a) RNH₂, AcOH, NaBH(OAc)₃, 1,2-dichloroethane; b) $R = Bn$: BnNH₂, MeOH, Ni-Raney (72%); c) R_1 = Boc: TFA, H₂O, then (Boc)₂O, TEA, DMAP, THF; d) R_1 = Bn: BnBr, DIPEA, MeOH (75%); e) IN HCl (100%); f) KCN, (NH₄)₂CO₃; g) R₁ = Boc: LiOH; h) R = R₁ = Bn: AcOH, HCl 37%, 50°C (77%); i) MeOH (61%); 1) CH₂Cl₂, BuSO₂Cl, pyridine (48%); m) H₂, MeOH, Pd/C, 5 bar, 50°C (38%)

Scheme 27

was obtained, alkylated and hydrolyzed to 156b ($R_1 = Bn$, 75%). The Bucherer reaction and hydrolysis gave the N , N -cis-157b compound. Protection of the carboxy and amino groups as methyl ester and sulfonylamide, and reduction of the benzyl group formed compound 158, an important precursor for antithrombotic compounds. Other amides were reported.

1,4-Diamino-1,4-cyclohexanedicarboxylic acid 161 was used for the preparation of the 4-aminoethanthiol derivative 162 as antiradiation agent (Scheme 28).⁴⁵ Compound 161 was obtained from 1,4-cyclohexanedione 159 (Meth. A) which was converted (Strecker reaction) into 1,4-diamino-1,4-cyclohexanedicarbonitrile 160 and hydrolyzed. Yield was not given.

3. Oxygen $Ac_{s}c$ Compounds

As far as oxygen-containing substituents are concerned it is obvious that cyclohexene and cyclohexane-1-amino-1-carboxylic acids bearing one or more free OH groups are the most interesting compounds of this family and numerous other derivatives have been prepared in view of their transformation into them. However, it should be considered that the final step (mainly a deprotection step) is not always described, notwithstanding it should be considered possible in principle. Compounds having from one to four OH groups are known. In the following text monohydroxylated compound are treated separately from polyhydroxylated ones. Table 1 offers an overview for easier location of items. The three main methodologies for their preparation are Meth. C, A, D. It appears that one most useful way to construct the cyclohexane skeleton is by

the Diels-Alder cycloaddition reaction. Good possibilities for the introduction of precursor groups of the OH function and carbon containing ring or chain **are** given and steric control is satisfactory in many cases. Functionalized or not functionalized dienes have been used and **as** dienophyles **alkylidene-imidazolones, methylene-oxazolidinones** and, far more interestingly, alkylidene-oxazolones or substituted 2-aminoacrylates have been extensively employed for generating monocyclic or bicyclic precursors. Of importance, both in the case of oxazolones and in the case of acrylates is the presence of a chiral substituent, which can induce chirality in the cycloaddition product and ultimately in the final hydroxy amino acid. A case of chiral center on the oxazolidinones is also known.

An interesting and widely used possibility is offered by the versatility of the double bond in the cyclohexene derivatives directly obtained from Diels-Alder cycloadditions. Besides traditional reactivity, cases of intramolecular interaction with reactive groups have been described which allow the introduction of further hydroxy groups with high regio- and stereocontrol (iodo-oxazination and lactonization reactions). Methodology A was applied when an oxygen functionalized cyclohexanone was used but in this case the availability of other substituents on the ring is limited. Moreover, this synthetic approach is very good when chiral syntheses were requested. The ring closing metathesis reaction (Meth. D) offers the possibility to obtain both β - or γ -hydroxy unsaturated Ac_{ϵ}c compounds in the chiral form but in general with low diastereoselection. Furthermore, the overall yields of amino acid **are** generally low.

i. Monosubstituted compounds

In the following chapters the preparation of different monohydroxy amino acids was reported according to the three different methodologies reported before. The use of Diels-Alder reaction allowed to introduce the oxygen atom both in β and δ position. The use of heterofunctionalized dienes like 1 -acetoxybutadiene **(136~). 1-acetoxy-3-methylbutadiene (136d), 1** -tributylsilyloxybutadiene **(136e),** 1-methoxybutadiene **(136g)** or **1-trimethylsilyloxybutadiene (1%)** ensures the functionalization of the P-position of the cyclohexyl ring. Instead, Danishefsky's diene (136f) or 2-methoxybutadiene (136i) ensure the δ -functionalization. The first examples of Meth. C, albeit scarcely detailed, consisted in the use of imidazolones and oxazolones **as** dienophiles.

The use of imidazolone **163** and 1-acetoxybutadiene **(136c)** or I-acetoxy-3-methylbutadiene **(136d)** (Meth. C) at high temperature formed a mixture of isomeric and regioisomeric compounds in poor yield **(1550%).** *Scheme* 29 describes the major regioisomeric compounds 164 characterized by a β -acetoxy substitution.⁴⁶ The cycloaddition reaction of 4-ylideneoxazolone **Z-165a** with dienes was first reported in 1989.47 The reaction was done with dienes **136c,e,f** at high temperature (Meth. **C)** giving the corresponding cycloadducts **166a-c** in *50-* **100%** yield *(Scheme* 29). Both for compounds **164** and **166** the configuration was not assigned.

According to Meth. C, three different synthetic approaches were adopted for the preparation of the *N.O-cis* hydroxy compounds. The first one,³⁸ reported in *Scheme 24*, takes advantage of the use of both β -chlorocycloadducts **137a** and **138a**. In fact, a single diastereomer **143a** was obtained using oxazolines 142 as key intermediates (Meth. A). As expected, the *N,Cl-trans* cycloadduct **138a** or the corresponding amide derivative **141** on reaction with methanol in the presence of an acid or basic catalyst, respectively, were transformed into the oxazoline derivatives **142a** (80%) or **142b** (91%). The use of the diastereomeric amide **N,CI-cis-140** under basic conditions gave the same oxazoline **142b** (95%). For its formation it was assumed the anchimeric assistance of the amid0 group. The hydrolysis of oxazolines **142** gave the amino acid **143a.** *An* improvement of yields in compound **143a** was observed **starting** directly from oxazolone **Z-12i 48** *(Scheme3O).* Oxazolone **12i** having an ethoxycarbonyloxy group at the double bond, is the sole

a) 136a or 136b, EtAlCl₂, CH₂Cl₂ (70%); b) 168: EtOH, H⁺ (65-70%); 169: THF, HCl (68-60%); 170: NaOH, EtOH, then H⁺ (68%); 143: Na₂O₂, H₂O, 50°C (74-78%); c) NIS, CH₂Cl₂; d) Bu₃SnH, CH₂Cl₂; **e**) **HCI**, reflux (92%); **f**) NaHCO₃, I_2/I , **EtOH**; **g**) NaOH, THF/H₂O, then H⁺; **h**) R = Me: Na₂O₂, H₂O, **SOT; i) R** = **H: 20% HCI, reflux Scheme 30**

heterosubstituted dienophile which was used in a Diels-Alder reaction for the introduction of the hydroxy group in the β-position (Meth. C). The diastereoselective synthesis of N,O-cis-βhydroxy unsaturated Ac,c compounds **143a,b** was reported by reaction of **2-12i** with **2,3-** dimethylbutadiene (136a) and 2-methylbutadiene (136b). The reaction was catalyzed by ethylaluminum dichloride and gave a single cycloadduct 167 which was selectively deprotected to compounds 168-170 or to the free amino acids 143 (74-78%).

The third approach takes advantage of the use of the acrylate 177a and diene 136f but suffers from many stages and low overall yield.⁴⁹ (Scheme 31) The N,O-cis saturated isomer 76a (25% from 177a) was obtained from cycloadduct 178a transformed into unsatutated ketone 179a

a) 136f, dioxane, reflux; b) NaF, THF, then column chromatography (48%); c) 1,3-propanedithiol,
BF₃•Et₂O, CH₂Cl₂ (77%); d) Ni-Raney, EtOH (62%); e) 136g, toluene, 85°C (72%); f) H₂Pd-C,
MeOH or CH₂Cl₂ (78-99 R_1 = Ph: -78°C (98%); m) MsCl, TEA, CH₂Cl₂ (95%); n) TFA, THF, 80°C (95%); o) AcCl, TEA, CH_2Cl_2 (80%); p) PhCSCI, DMAP, MeCN, then Bu₃SnH, AIBN, toluene (70%)

Scheme 31

after methoxy group elimination. The key intermediate oxazoline **18Oa** allowed to introduce a poxygen atom on the ring. The reduction of the keto group in **180a** to hydroxy followed by water elimination produced **181.** Attempts to reduce the double bond failed; instead, the hydrolysis of the oxazoline ring gave the amino acid derivative **182** which was first protected at the nitrogen atom and then hydrogenated and hydrolyzed to racemic amino acids **76a.** The synthesis of the **6** phenyl-substituted *N,O-cis* derivative **185,'O** *(50%* overall yield) *(Scheme* 31) was reported by the same authors from oxazoline **180b** obtained from **179b** (see *Scheme 34* for its preparation). In this case the elimination of the hydroxy group was done by reduction of the ketone to alcohol **183** from which oxazoline **184** was obtained by activation of the hydroxy group and reduction. Compound **184** was hydrolyzed to **185.** The **1** -aminoacrylate **177a** was also used for the preparation of the *N,O-trans* 2-hydroxy Ac_cc compounds **76b**⁴⁹ by reaction with dienes **136f** or **136g** (Meth. C) (Scheme 31). Cycloadducts **178a** and **187** were obtained, respectively (only small amounts of the *cis* isomer were detected (10%)). The keto group in **178a** (R = Ph) was reduced, *via* thioacetal **186,** to compound **188** (22% overall yield). Better yields **(59%** from **177a)** in the preparation of **76b** were achieved when the same compound 188 was obtained from **187** by reduction of the double bond. The hydrolysis of **188** formed the racemic amino acid **76b.** Resolution of both racemic **trans-188** and **cis-182** was done with L-phenylalanine and S-2-acetoxypropionyl chloride, respectively.

Three chiral dienophiles were used in the Diels-Alder reaction: the acrylate **177c** and the oxazolone **165b** having a chiral dioxolane residue on the methylenic carbon and the oxazolidinone **195** having a chiral center on the ring. The chiral synthesis of the β -hydroxy Ac_{ϵ}c compounds **191** and **193** *(Scheme* 32) was reported from **177c** and oxazolone **16%** respectively,

a) **136h** ,CH2Cl2, **14 Kbar,** 60°C (70%); b) **136h,** CH2CI2, 25°C **(70%);** *c)* **136h,** CH2C12, 12.5 **kbar,** 25°C (90%); d) citric acid, MeOH, then MnO₂, CHCl₃ (60%)

Scheme 32

using 1-TMSO-butadiene (136h) (Meth. C).⁵¹ The acrylate worked at 60° and 14kbar giving a mixture of two diastereomeric *syn* compounds **endo-191, em-191** and the *anti* derivative **192** in 6:3:1 ratio (70%). Better results were observed with the more reactive oxazolone **16%** which reacted at room temperature giving two *syn* cycloadducts endo-193 and exo-193 (1:2; 70%). After deprotection at the oxygen atom and oxidation, they were transformed into the chiral 2 keto derivative 194. Starting from 165b and under pressure, only the pure diastereomer *N,Otrans* **endo-193 (90%)** was obtained. In all cases stereochemical control both at C-1 and C-6 (R,R configuration) was observed and the $2R$ isomer was favored with respect to the $2S$ one. In conclusion the use of oxazolone **16%** and diene **136h** is a **good** way for the chiral synthesis of N, O -trans β -hydroxy or β -oxo unsaturated Ac₆c compounds.

The chiral methyleneoxazolidinone **195** was **used as** dienophile in Diels-Alder reactions (Meth. C) and was less reactive than ylideneoxazolones (Scheme **33).52** When reacted with 1 methoxybutadiene **(136g)** the corresponding chiral cycloadducts **196a,b** (51%, 82:18) were

obtained at *60"* in 10 days. The cycloaddition reaction **was** regioselective both diastereomers having the *S* configuration at the spiro carbon. This synthetic approach is more general considering the possibility to eliminate the chiral residue by simple hydrolysis of the heterocyclic ring. The same compound 195 was reacted with diene 136f, which allowed to obtain a β , δ -oxygen substituted cycloadducts 197a,b (40%, 1:1) operating at 60° for 2 days. Also in this case a good regioselectivity and **S** control on spiro center was observed.

The stereochemistry of the cycloaddition of Danishefsky's diene (Meth. C) was detailed both starting from Z-phenylydeneoxazolone 165c⁵³ and chiral oxazolone 165b^{51,54} operating at 120" and room temperature, respectively *(Scheme* 34).

**a) 136f, toluene, 120°C; b) 136f, CH₂C1₂, 25°C; c) 0.005 N HCl/THF (1:4); d) MeOH/DBU; e) H₂, Pd/C, CH₂Cl₂ (100%);
f) 6N** HCl, then EtOH, propylene oxide; **g) 136i**, Me_2 CO/H₂O, reflux (65%); **h) NH₃, MeOH, then c (95%); i) OH(CH₂)₂OH, TsOH, reflux; 1) NBS, Hg(OAc)₂, MeOH (58%) Scheme 34**

Two diastereomeric cycloadducts *exo*-198 and *endo*-198 (R_1 = Ph) were obtained from 165c, and were hydrolyzed directly to the corresponding ketones *exo*-199 and *endo*-199 (R₁ = Ph; **93%,** 4555). Methoxy group elimination (179b), reduction of the double bond and hydrolysis gave ketone 200a *(55%* overall yield; cis-relationship between the amino and phenyl groups). The cycloaddition reaction of Danishefsky's diene (Meth. C) was applied also to a series of arylidene- and **heteroarylideneoxazolonesss** and to alkylideneoxazolones 165.56 Chiral oxazolone 165b afforded the diastereomeric compounds exo/endo-199 (1:1). They were transformed into the chiral unsaturated ketone 179 ($R_1 = \text{diaxolane}$) (48%,⁵¹ 76%⁵⁴). This was reduced to $1S,2R-201$ as depicted in *Scheme 34*. The methoxy group elimination is easier in the case of *endo* compound. Keto-amino acid 200b (trans-relationship between amino and phenyl) was obtained from methyl E-2-cyanocinnamate 202 and 136i (Meth. C) (Scheme 34). The cycloadduct 203 was obtained which, after ammonolysis of the ester group and hydrolysis of the enol (compound 204), protection of ketone and Hofmann transposition, gave 205 which was hydrolyzed to 200b *(66%).53*

Compound 189 (Scheme 31, from acrylate 177b and 136f; Meth. C), was used to prepare diastereomeric 4-hydroxy- **1** -amino acids 210 and 211 by reduction of the ketone group (Meth. A) (Scheme 35).⁵⁷ NaBH₄ favored the formation of the *N*, *O*-cis compound 206 (206/207,

a) NaBH₄, EtOH, 0°C (91%); b) L-Selectride, THF, -78°C (91%); c) R = Ac: Ac₂O, Sc(TfO)₃, MeCN (75%); d) R = Ms: MsCl, TEA, CH₂Cl₂, 40°C (68%); e) CsF, BzOH, DMF 80°C (68%); f) HCl 6N, reflux(70-78%). **Scheme 35**

 $80:20$). In contrast, L-selectride produced isomer 207 as the major one $(206/207, 10:90)$. The mixture of compounds 206.207 was O-acetylated $(208a.209a)$ or O-mesylated $(208b.209b)$ and separated. The hydrolysis of 208b and 209b gave the corresponding amino acids 210 and 211. Compound 208b was converted into the diastereomer 209c with CsF and benzoic acid. According to methodology A and using oxygen substituted cyclohexanones as starting materials different synthetic approaches to 1-amino-2-hydroxycyclohexanecarboxylic acids (76) were reported and all four stereomers were synthesized and isolated. The first example was the synthesis of an achiral diastereomeric mixture of the 2-hydroxy derivatives 76 according to the synthetic pathway depicted in Scheme 13 from ketone 74a through hydantoin 75a.⁵⁸ The enantioselective synthesis of the diastereomeric *IR.2S*- and *IR.2R*-1-amino-2-hydroxycyclohexanecarboxylic acids 76a and 76b (Scheme 14) was reported by Ohfune²⁵ (Meth. A) by transforming 1,2-cyclohexanediol into a mixture of the diaster eomeric ketones 83 (1:1, 70%), through intermediates 82. Following the synthetic scheme adopted for the analogous cyclopentyl derivatives, ketones 83 were transformed into a mixture of diastereomeric enantiopure amino acids 76a and 76b (4:1) in 64% overall yield. The four isomeric 1-amino-2-hydroxycyclohexanecarboxylic acids 76 were obtained by Fondaker⁵⁹ (Scheme 36) starting from the racemic ketone 74b (Meth. A) which was reacted with (S)-1-phenylethylamine (PEA) giving a mixture of E and Z isomers 212 (10:1). The reaction with trimethylsilyl cyanide produced four stereomers 213. Depending on the solvent, the N,O-trans/cis ratio was changed. Thus, in MeOH trans isomers 213 were favoured (trans/cis = 74:26), whereas the cis compounds 213 (trans/cis= $25:75$) was predominant in hexane. The hydrolysis of diastereomeric nitriles 213 gave a complex mixture which was separated giving a mixture of N, O -trans aminoamides 215a,b and the hydrogenolyzed $N, O\text{-cis}$ compound 214. The latter was hydrolyzed to amino acid $IS, 2R-76a$ (86%, 88% ee). Separation of diastereomers trans-215a,b followed by hydrogenolysis and hydrolysis of the major stereomer afforded the amino acid (1S,2S)-76b (89%, 98% ee). The authors also described the preparation and isolation of the corresponding enantiomeric acids $IR, 2S-76a$ and $IR, 2R-76b$ when the $(R)-1-$ PEA was used as the chiral auxiliary.

a) (S)-1-PEA toluene, 120°C (93%); **b**) **TMSCN**, ZnCl₂, MeOH or hexane (98-99%); **c**) **H**₂SO₄ from -10 to 0°C, then silica gel; d) HCl 12N; e) LPLC: Lobar, then H₂ Pd/C, EtOH, 45°C, 5 bar **Scheme 36**

Two synthetic procedures to β -oxo Ac₆c compounds were reported the first one being more convenient (Meth. A) in term of starting material, yield and stereocontrol *(Scheme* **37).60** In fact, starting from the chiral β -enamino ester 216 and ethyl N-[(4-nitrobenzenesulphonyl)oxy]carbonate (through aziridine **B)** compound **93c** was obtained which was transformed into the

corresponding ketal **217** *(95%,* 60% *d.e.).* Alternatively, using the same synthetic procedure applied to the Ac₅c ring (Meth. A) *(Scheme 15)*, β -ketoester **91b** or its silyl enolate were trasformed into **2-0x0** derivative **93b** (60-70%) through intermediate **92b.20** The synthesis of benzocondensed derivatives was also reported. Methodology D was used for the preparation both of **C-P** and C-y amino acids starting from chiral bislactims having a hydroxy functionalized chain *(Scheme 38).*²⁶ The reaction is general for the preparation of β -hydroxy Ac₅c, Ac₅c and Ac₇c, but in the first case a different patway was followed. By reacting **218a-c,** with acrolein, two diastereomeric hydroxy derivatives **219a-c** and **220a-c** were obtained in about 1:2 ratio. From compounds **219b,c** and **220b,c** spiro compounds **221a,b (a: 89%, b: 88%)** and **222a,b (a:** *59%,* **b: 63%)** were obtained, respectively. Formation of the five-membered spiro-ring failed. Hydrolytic cleavage of **221a** and **222a** gave **223** and **224,** respectively, in **81** and 60% yield. Amino ester **225** was obtained in 66% yield from **221b. By** contrast, in the case of **222b.** a mixture of hydrolysis compounds **226a-c** was found **(65%).** Bislactims **220b,c** were also used for

a) BuLi, THF, CH₂CHCHO, -78°C (63-70%); b) 2% Ru(II), benzene 20°C; c) 2% Ru(II), DEC, 222: reflux, 228a: 40°C, 228b: 80°C; d) 0.2M TFA, MeCN; e) (COCl)₂, DMSO, CH₂Cl₂, -60 to -10°C (71-73%); f) Ac₂O, DMAP, CH₂Cl₂ (90-93%)

Scheme 38

the preparation of β -ketoamino esters 229,230.⁶¹ The oxidation of the hydroxy group produced 227a,b transformed into the corresponding spiro-compounds $228a(69%)$ and $228b(28%)$. The hydrolysis of these latter resulted in low yield giving ketones 229 (37%) and 230 (47%), respectively. For the preparation of y-hydroxy Ac₆c compounds the pathway depicted in Scheme 39 was followed (Meth. D).⁶² Starting from the bislactims **234a** and **235a** (38% yield from reaction of compound 233a with vinylmagnesium bromide) the corresponding enenantiopure derivatives 236a (73%) and 237a (75%) were isolated which were transformed into the unsaturated amino esters 238 and 239 in very low yields (38, 19%, respectively) having the N,O-trans and N,O-cis relationships, respectively. The saturated 3-hydroxy $Ac_{c}c$ compounds functionalized with a methyl group at $C-5^{63}$ (*Scheme 39*), were obtained from bislactims 234a and 235a (mixture of compounds) by oxidation to the keto derivative 242 which was transformed, using ring closing methatesis reaction, into spiro cyclohexenone 243 (37% overall yield). It was possible to increase the ketone yield (52%) starting from the cyclic compounds $236a,237a$ by the Swern oxidation.

The unsaturated ketone 243 was reacted with lithium dimethylcuprate giving a single diastereomer 244 which was reduced to a mixture of alcohols 245 and 246 (37 and 57%, respectively). Mild hydrolysis gave the amino acid 247 (38%) and the protected one 248 (77%), respectively.

An alternative and more efficient way (better yield and diastereocontrol) to y-hydroxy compounds is shown in *Scheme 40*. The 1-amino-3,4-cyclohexenecarboxylic acid derivatives 249a,b, obtained through Diels-Alder reaction and having an amidated nitrogen atom, were the starting materials for the regio- and stereocontrolled functionalization of C-3 on the cyclohexyl ring with an hydroxy group (Meth. A). The diastereoselective synthesis of compound 252a (Scheme 40) was reported⁶⁴ using the iodo-oxazination reaction which allowed to control the cis relationship between the oxygen and nitrogen atoms. Oxazine 250a was obtained from

a) $I_2/dioxane$; b) Bu_3SnH , CH_2Cl_2 , 60°C; c) TFA, THF/H₂O, 80°C; d) KOH, EtOH, then H⁺; e) NIS, CH₂Cl₂; f) 6N HCl; g) H₂O; h) NBS, Hg(OAc)₂, DMF; i) 3N HCl

Scheme 40

cycloadduct 249a and was transformed into 251a by iodine reduction and hydrolytic cleavage. Hydrolysis of 251a gave amino acid 252a. The chiral IS, 3R-hydroxyamino acid 252b (53% overall yield) was obatined by the same authors⁶⁵ from the chiral cycloadduct 249b using, in the iodo-oxazination reaction, NIS instead of iodine (intermediate 250b) (Scheme 40). A way to chiral compounds $N, O\text{-cis-}252a^{66}$ is from the chiral cycloadduct 1S, 2R-253, taking advantage of the presence of a carboxamido group (Scheme 40). Compound 253 was reacted with iodine giving the intermediate 254. After removal of the iodine atom and hydrolysis (intermediate 255), Hoffman transposition and hydrolysis, the chiral amino acid $IR, 3R, 5R-252a$ was isolated in 70% overall yield. Enantiomer 1S, 3S, 5S-252a was also reported.

4-Hydroxy Ac_sc compounds were prepared from 4-benzoyloxycyclohexanone 256 using Bucherer-Berg reaction (Meth. A) (Scheme 41).⁶⁷ A mixture of two diastereomeric hydantoins 257a and 257b was obtained in favor of the N, O -trans derivative (cis/trans, 1:3). They

Scheme 41

were separated and transformed into the corresponding amino acids 210 and 211. Alternatively, the cis-isomer 257a was selectively prepared by treating 256 with KCN/NH₄Cl and cyclizing the amino nitrile with KOCN. This method compared to those reported in Scheme 35 appears less diasteroselective and versatile in term of revertion of diasteroselection.

The synthesis of 4-oxo Ac₆c compounds (Scheme 42) was made via three-component reaction of 1,4-cyclohexanone monoacetal 154, phenylalanine as nitrogen donor and an

a) L-Phe, RNC, MeOH, -78 to 25°C (80-88%); b) MeOH, Pd(OH)₂, H₂ (58-70%); c) 6N HCl, reflux (60%); d) KCN, NH₃; e) (NH₄)₂CO₃; f) Ba(OH)₂; g) NaCN, (NH₄)₂CO₃, H₂O, 60°C; h) 0.5 N NaOH, reflux; i) ArCH₂COCl, TEA, THF, 0°C; 1) CH₂Cl₂, H₂SO₄, MeOH, 40°C

Scheme 42

isocyanide (Meth. A).68 The reaction formed compound **258** which was deprotected to **259** and hydrolyzed giving **26Oa.** A way to amino acid **26Ob** *(Scheme 42),* protected at the keto moiety, was reported from **154** (Meth. A). The Strecker reaction afforded nitrile **261,** then transformed into hydantoin **262.** The hydrolysis of the heterocyclic **ring** under basic conditions gave amino acid 260b.⁶⁹ Alternatively,⁷⁰ compound 260b was obtained by hydrolysis with sodium hydroxide of the hydantoin **262** directly obtained from acetal **154** with sodium cyanide and ammonium carbonate. Yields were not reported. The 4-keto amino acid derivatives 264⁷¹ (Scheme 42) (used as insecticide) were prepared from aminonitrile **261** which was acylated to compounds **263.** The nitrile group was transformed into ester 264 by reacting with CH₂Cl₃, H₂SO₄ and methanol. By an analogous procedure, the synthesis of 4-methoxy derivatives was reported.

ii. Polysubstituted compounds

Several dihydroxy-Ac,c derivatives were reported, *i.e.* the 2,4-,2,5-, 2,6- and 3,4-derivatives. In most cases the above described amino acidic compounds were used for further transformations. The synthesis of the 2,4-dihydroxy derivative 190⁵⁰ (Meth. A) *(Scheme 31)* was reported from oxazoline **18Ob.** The keto group was selectively reduced giving the *0,O-cis* dihydroxy compound **183,** transformed into amino acid **190** (51% overall yield).

Regio- and diastereoselective syntheses of 2,5-dihydroxy derivatives were reported from the 2-hydroxy derivatives **168** and **169** (Meth. A) *(Scheme 30).48* The *cis* relationship between the hydroxy groups was ensured starting from esters 168a,b which, through the iodooxazination reaction followed by reduction of iodine with tributyltin hydride, were transformed into **171a,b** in 70-75% overall yields. *On* reaction with HCI, compound **171b** was converted into the expected dihydroxy derivative **172b** (92%), but **171a** gave amino acid **143a.** To ensure the *trans* relationship between the hydroxy groups, the iodo-lactonization reaction was applied to acids **169a,b.** According to *Scheme 30,* these compounds were transformed into lactones **173a** (41%) and **173b**. From **169b** a second regioisomer **174** was obtained **(173b/174, T**= 25° , 1:2; T= 45", 1:3.5). The hydrolysis of both the lactone and benzamido groups in compounds **173** and **174,** afforded 2,5- and 2,4-dihydroxyamino acids **175a** (67%), **175b** (89%) and **176** (56%), respectively. As for the corresponding Ac,c derivatives, the methodology D was used for the preparation of the chiral *(1 S,3R,4S)-* 1 **-amino-3,4-dihydroxycyclohexanecarboxylic** acid derivative 109 (62% overall yield) from functionalized lactim 105b through the dihydroxy derivative **107b,** which was the major isomer **(106b/107b,** 6:82) *(Scheme 28).29* In contrast to the Ac,c analogue, the *N,O-trans* relationship was preferred in this case.

2,6-Dihydroxy Ac₆c compounds 266^{72} (*Scheme 43*) were prepared by a condensation reaction of glutaraldehyde and nitroacetate (intermediate **265)** (Meth. B) and reduction and hydrolysis. The *trans* relationship between acetate **and** acetamido groups was proposed for compound 267. From (-)-quinic acid 268⁷³ the chiral 1-amino-3,4,5-trihydroxycyclohexanecarboxylic acid **270** was obtained according **to** *Scheme 44* (Meth. A). Compound **268** was transformed into the lactone 269a which was tosylated to compound 269b. This was transformed into **270** by reaction with ammonia and hydrolysis.

The enantioselective synthesis (Meth. B) *(Scheme 45)* of 1 **-amino-2,3,4,5-tetrahydroxy**cyclohexanecarboxylic acid **274** was reported from the azidolactone **272,** obtained from **271,** tion gave the corresponding amino acid 274 in 56% yield.^{31b}

4. Sulfir Ac,c Compounds

According to synthetic *Scheme* 21, the 2-thiobenzyl derivative **119b** was synthesized **as** described for the 2-thio Ac,c analogue.33 The 3-methylthio derivatives **122b** (stereomeric mixture) were obtained *(Scheme 21)* from 2-cyclohexenone (Meth. A) through the nitrile intermediate **121b (36%** overall yield).34

5. Phosphorus Ac,c Compounds

The diastereoselective synthesis of the 3-phosphino Ac_c compound 277,⁷⁴ an inhibitor of glutamine synthetase, was reported. Reaction of 2-cyclohexenone with methyl phosphonite afforded **275.** It was transformed into spiro hydantoin derivatives **276** (9223). The major isomer is indicated in Scheme 46. The hydrolysis of **276** gave the amino acid derivative **277,** having *N,P-trans* relationship, in 96% *d.e.*

a) MeP(OEt)₂, EtOH; b) 6N HCl, reflux; c) KCN, (NH₄)₂CO₃, EtOH/H₂O, 55°C (73%); d) Ba(OH)₂, **H20,** reflux, then Dowex *50* (NH40H). then NaOH (60%)

Scheme *46*

Two diastereomeric bioactive 3-phosphono derivatives **135a** and **135b** (**1 :4)** were synthesized according to *Scheme 23*, already discussed for the $Ac_{\mathcal{S}}c$ analogues. The amino nitrile derivatives **132a** and **132b** were obtained in 81% yield and hydrolyzed to compounds 135.³⁶

VI. 1-AMINOCYCLOHEPTANE-1-CARBOXYLIC ACIDS

Few examples of heterosubstituted Ac_{-c} compounds were reported, all of them having an hydroxy group as substituent. For their preparation *i)* the RCM reaction (Meth. D) and *ii)* the 1,3-dipolar cycloaddition reaction (Meth. C) were used.

1. Oxygen Ac,c Compounds

The preparation of unsaturated β -hydroxy- and γ -hydroxy Ac₇c compounds was reported using the RCM reaction (Meth. D). Both diastereomeric β -hydroxy amino esters 225 and 226²⁶ and the corresponding ketone 230⁶¹ were prepared in satisfactory yield according to the pathways depicted in Scheme 38 (see analogous $Ac_{\alpha}c$ syntheses for details). According to the synthetic pathway depicted in *Scheme 39*, both diastereomeric unsaturated γ -hydroxy Ac_,c compounds were obtained from6' bislactims **234b** and **23%** deriving from **233b** (36 and **43%** yield, respectively). The RCM reaction did not work when applied directly to hydroxy compounds which were transformed into the acetoxy derivatives **234c** and **23%.** *Good* yields were obtained when these latter were cyclized giving *236b* and *237b.* After hydrolysis the unsaturated amino esters **240** and **241** were obtained in 48 and 30% yield, respectively, having the *N,O-truns* and *N,O-cis* relationships, respectively, **as** demonstrated by the lactone structure of compound *240.* The preparation of 4-hydroxy Ac,c compound **282** was done by an intramolecular cycloaddition reaction of nitrone **279a** functionalized with an ally1 chain (Meth. C) (Scheme *47).'5*

The reaction afforded cycloadduct 280 transformed, by hydrogenolysis, into **281.The** catalytic hydrogenation of **281** under high pressure led to the cycloheptane amino ester **282** with *N, 0-cis* relationship. Using the same synthetic strategy, 4-hydroxycyclooctyl- **283** and 5-hydroxycyclononyl- 1 -amino esters **284** were obtained from **279b,c,** respectively.

VII. 2-AMINOBICYCLO[3.1.0]HEXANE-2,6-DICARBOXYLIC ACIDS

A large number of heterosubstituted 2-aminobicyclo[3.1 **.O]hexane-2,6-dicarboxylic** acid derivatives were reported because of their interest as potent agonists on the aminoacidic receptors in the CNS system. As heteroatoms, F, N, 0, **S** and P were **linked** at the C-5 ring both at C-3 and/or C-4. The preparation of a series of fluoro-substituted compounds at C-3 ring was also reported. The main synthetic approaches to these compounds were i) the cyclocondensation reaction giving a heterofunctionalized bicyclo^[3.1.0]hexane ring (Meth. B) and *ii*) the cyclopropanation reaction of cyclopentenone (Meth. C) followed by functionalization of the cyclopentyl ring. In this case the ethyl 2-0x0 carboxylate **285,** obtained from cyclopropanation reaction of 2-cyclopentenone, was the common starting material for the preparation of a large number of heterosubstituted $Ac_{[3,1,0]}c$ compounds.

1. Halo **AC,~,,.~,~** *Compounds*

Several **fluoro-2-aminobicyclo[3.1.0]hexane-2,6-dicarboxylic** acids *(i.e.* 3-fluoro-, **4** fluoro-, 3,3-difluoro-, 6-fluoro-, 6-fluoro-4-hydroxy- and 6-fluoro-40x0) were prepared and in some cases both diastereoselective and enantioselective syntheses were reported. Compound **285** was used to prepare 3-fluoroamino acids **292-2p476** (Scheme *48).* a-Fluorination of ketone **285** produced a mixture of the bis-fluorinated compound **288** (27%) and of two diastereomeric compounds 286 and **287** (50%, 1:3). These latter were reacted under Bucherer-Berg conditions to yield three hydantoins **289,290** and **291** which were hydrolyzed to the corresponding amino acids **292,293** and **294.** The racemic hydantoin **289** was resolved and each enantiomer gave (+)-

or *(-)-292.* Starting from difluoro intermediate **288,** the 3,3-diflUOrO amino acid *2%* (26%) was obtained through hydantoin *295.* A different and more efficient approach for the chiral synthesis of compound *N,F-cis (+)-292* was followed from chiral ketone *285,* which was transformed into the unsaturated ketone *(-)-2W.*

a) $EtO_2CCH_2SMe_2$ \cdot Br, DBU, PhMe; b) LHMDS, TMSCI, THF, then $(PhSO_2)_2NF$, $CH_2Cl_2 (50%)$. c) (NH4)2CO3. KCN, EtOWH20; d) 60% H2S04; **e)** *2.5* or 3M NaOH, H20; **f)** LHMDS, **TMSCI,** THF, then Pd(OAc)z, MeCN (86%); **g)** TBHP, Triton B, PhMe **(73%);** h) **KF-HF,** ethylene glycol (46%); **1)** H2, PdC, EtOH **(75%) Scheme** *⁴⁸*

The epoxidation reaction gave a single diastereomer **(+)-298** which, on reaction with potassium fluoride in ethylene glycol, gave two unsaturated fluorinated esters (-)-299a,b (18:28). Reduction of the double bond followed by Bucherer-Berg reaction and hydrolysis afforded *lS,2R,3S,SR,SS-(+)-292* (48%). For the preparation of *&N,F-cis* fluoroamino acids **303,307** and **31076** a different approach was followed (Meth. B) (Scheme *49).* In fact, starting **from** ethyl 2-2 **fluoro-5-carboxy-2-pentenoate** 300, an intramolecular reaction using **Cu(TBS),** afforded compound 301 which was transformed into the amino acid *IR*,2S*,5R*.6R*-303* (34%) *via*

a) (COCl)₂, hexane, reflux, then CH₂N₂, Et₂O, then Cu(TBS)₂, benzene, reflux (27%); b) IN NaOH; c) (NH&C03, KCN, EtOH/HzO (99%); d) **60%** HzS04, 150°C; e) LHMDS, TMSCI, THF, then Pd(OAc)₂, MeCN (89%); f) **TBHP**, Triton B, PhMe (98%); g) (PhSe)₂, NaBH₄, AcOH, EtOH (76%); h) EDCHCl, EtOH, DMAP, DMF; i) TBDSMCl, imidazole, DMF, then HS(CH₂)₂SH, BF3mEt20, CHCI1; **I)** DMSO, DCC, Py-TFA **(73%)**

Scheme 49

hydantoin **302.** The 4-hydroxy and 4-0x0-amino acids **1** *R *,2S*,4S*,5S*,* **6S*-307** and **lR*,2S*,5S*,6S*-310** were prepared from **301** by way of epoxide **305** and unsaturated ketone **304.** The epoxide was regioselectively reduced to 4-hydroxy ketone **306** in which the fluorine and oxygen atoms are *trans.* According to the usual synthetic procedure reported before, compound **306** was transformed into amino acid *307* in very low yield (3%). The protection of the keto group and the oxidation of the hydroxy group in compound **306** gave intermediate **308** (48%), the precursor of the 4-keto-amino acid **310.** The synthesis of (+)- and **(-)-310** was achieved from chiral hydantoins **309.** From resolved pure enantiomers of 6-fluoroketone **301** both the syntheses of chiral (+)- and **(-)-303** and **(+)-310** were reported following the above described procedures. According to *Scheme 51* (see p. 186), the $1S^*$, $2R^*$, $4S^*$, $5S^*$, $6S^*$ -4-fluoro-2-amino acid **331** was obtained by reacting the 4-hydroxy compound **324a** with diethylaminosulfur trifluoride.⁷⁷

2. *Nitrogen Acl,,,.,,c Compounds*

Amino-(2,3- and 2,4-derivatives), azido- and nitro-substituted amino acids were reported. The preparation of chiral **(lS,2R,3S,5R,6S)-2,3-diamhobicyclo[3.1** .O]hexane-2,6-dicarboxylic acid **31878** (Scheme *50)* was made from hydroxy compound **315** (see below **for** its preparation) which was transformed into triflate 316. The reaction with sodium azide formed the diazido derivative 317 as single diastereomer which was reduced and hydrolyzed to the diamino

a) DIPEA, THF, BuLi, N-phenyl-bis(trifluoromethylsulfonyl)imine, -78 to 25°C (87%); b) Pd(OAc)₂, Ph₃P, BnOH, DMF, CO (75%); c) K₂[OsO₂(OH)₄], (DHQD)₂PHAL, K₃Fe(CN)₆, K₂CO₃, MeSO₂NH₂, tBuOH, H₂O (46%, ee 63%; 26%, ee 99%); d) SOCl₂, CH₂Cl₂, 40°C, then NaIO₄, RuCl₃•H₂O (98%); e) NaN₃, H₂O, acetone, 50°C (62%); f) (TfO)₂O, Py, CH₂Cl₂, -78°C (86%); g) NaN₃, DMF, 80°C (49%); h) AcOH, H₂O, H₂, Pd/C, then 10% HCl (87%); i) CH₂Cl₂, PCC, 0°C (78%); 1) thioacetic acid, 70°C, then LiBH₄, EtOH, H₂O, -50°C (88%), then 10% HCl (70%)

Scheme 50

acid 318 in which the two nitrogen atoms are cis. Diastereoselective synthesis of $(IR^*, 2R^*, 4S^*, 5S^*, 6R^*)$ -4-azido-, -4-amino- and -4-nitro-2-aminobicyclo[3.1.0] hexane-2.6dicarboxylic acids was reported (Scheme 51) from tosylated 324b. The azido derivative 325a was obtained by reacting the tosylate with azide ion. The reduction of the azido group gave the amino compound 325b. Several 4-amino derivatives were also reported by reacting amine 325b with the appropriate reagent.⁷⁷ The oxidation reaction of the amino group to nitro group with peracid gave 326 in poor yield. Finally, the deprotection of amino and carboxy groups gave the 4-nitro-amino acid 327 (Scheme 51).⁷⁹ The 4-hydroxyimino compounds 329 (E/Z , 2:1) were obtained from ketone 328 on reaction with hydroxylamine and hydrolysis.⁷⁷

3. Oxygen $Ac_{(3,1,0)}c$ Compounds

Two diastereomeric 3-hydroxy, the N,O-trans 4-hydroxy and the corresponding 3- and 4-cheto derivatives were synthesized. Furthermore, the chiral 3,4-dihydroxy compounds were reported. The synthesis of diastereomeric chiral (1S,2R,3R,5R,6S)- and (1S,2R,3S,5R,6S)-3-hydroxyamino

a) TMSCl, TEA, CH₂Cl₂, 0°C then Pd(OAc)₂, MeCN (80%); b) TBHP, DBU, THF, 0°C (89%); c) N-acethyl-L-cysteine, Na₂B₄O₇, (PhSe)₂, H₂O, EtOH, THF (82%); d) (NH₄)₂CO₃, KCN, EtOH/H₂O; e) 2M NaOH, then EtOH, SOCl₂, 0°C, then NaHCO₃, THF, Boc₂O (26% from 323a); (f) TsCl, Py (91%); g) NaN₃, DMSO, 35°C (98%); h) Ph₃P, THF/H₂O (62%); i) mClPBA, CHCl₃ (9%); 1) EtOAc, HCl, 0°C, then 1N NaOH (327: 71%; 330: 31%); m) DAST, CH₂Cl₂; n) PDC, CH_2Cl_2 (98%); o) AcOEt, HCl, 0°C, then NH₂OH HCl, AcONa, EtOH/H₂O, 80°C (84%), then 1N NaOH/THF (60-68%) p) IN NaOH (90%); q) method d then DMF, NaHCO₃, BnBr, 100°C (19%); r) Jones' reagent, acetone, 0°C (88%)

Scheme 51

acids was done using an original synthetic stategy starting from 285 (Scheme $50⁷⁸$ whose enol triflate 311 was transformed into unsaturated carboxylate 312. Asymmetric dihydroxylation gave the chiral cis-diol 313 which was protected at oxygen atoms. By reacting sulphate 314 with sodium azide compound 315 was obtained which was reduced and hydrolyzed to chiral amino acid 319 having N,O-trans atoms. By contrast, when 315 was oxidized to ketone 321 and then reduced and hydrolyzed. A mixture of the two diastereomeric alcohols 319 and 320 (1:1) was obtained. The diastereoselective synthesis of $(IR^*, 2R^*, 4R^*, 5S^*, 6S^*)$ -4-hydroxy-2aminop. **1.0]cyclohexane-2,6-dicarboxylic** acid **33277** was reported from hydroxy ketone **322** obtained by regioselective reduction of epoxide **298** *(Scheme 51).* The hydroxyketone **322** was trasformed into a **mixture** of hydantoins **(1:l).** The desired diastereomer **323a** was isolated by HPLC **(26%)** or by crystallization (8%) and hydrolyzed to **332** (90%). Hydantoin **323a** was transformed into protected 4-hydroxyamino acid **3%.** The synthesis of the methoxy derivative was also reported. Two different synthetic approaches to 4-oxoamino acids were reported *(Scheme 51).77* The first gave poor yield (19%) and was done by transformation of hydroxyketone **322** into benzylated hydantoin **323b** oxidized to ketone **333.** Deprotection was not attempted. The second route resulted in the 4-0x0-amino acid **330** starting from protected amino acid **3%** which **was** oxidized to ketone **328** and then hydrolyzed.

The enantioselective synthesis of **lS,2R,3S,4R,5R,6R-3,4-dihydroxy-2-aminobi**cyclo[3.1 **.O]hexane-2,6-dicarboxylic** acid **341** was performed from the chiral unsaturated ketone **334** *(Scheme 52).*O*

e) DBU, 18-Crown-6, NaN3, MeOH (84%): *0* Hz, PdC, AcOEt, 30 psi **(71%); g)** MeCOCI, TEA, CH2C12, then TFA, H2O (67%) **Scheme ⁵²**

The cyclopropanation reaction of **334** (Meth. **C) occurred** in quantitative yield **and** gave a single diastereomer **335.** The Bucherer-Berg and Strecker reactions resulted in the hydantoin **336** and nitrile **337,** as pure isomers, having the amino group *cis* to the hydroxy ones. By contrast, using a modified Corey-Link reaction, it was possible to prepare the amino acid derivative **341** having the opposite configuration at C-2. On reacting **335** with CHC1, in the presence of the base, compound 338 was obtained and was rearranged to 339 by sodium azide in basic conditions. Reduction of azido group to **340,** protection of amino group and deprotection of hydroxy groups afforded the chiral amino acid derivative **341.**

4. *Sulfur Ac_{13,1,01}c Compounds*

A series of *(lS*,2R*,4S*,SS*,6S*)-4-thioaryl-* or **-4-thioalkyl-2-aminobicyclo[3.1** *.O]* hexane-2,6-dicarboxylic acids **343** were prepared as depicted in Scheme **53 from** intermediate **297** which was reacted with mercaptans giving **342** in good yield. **342** was transformed into amino acids **343** *via* the Bucherer-Berg reaction. Oxidation to sulfoxide and sulfone was also described.⁸¹

5. Phosphorus AcI3.,.,,c Compounds

Using the Strecker reaction in the presence of alumina and ultrasound, the **1S*,2R*,4R*,5S*,6S*-2-aminobicyclo[3.1.0]hexane-2,6-dicarboxy-4-phosphonic** acid **345** was prepared from intermediate 344 obtained by reacting 297 with triethyl phosphite (Scheme 53).⁷⁷

a) RSH, THF, TEA; b) (NH₄)₂CO₃, KCN, EtOH/H₂O; c) 2N NaOH; d) (EtO)₃P, PhOH, 100°C **(92%);** *e)* **KCN, NH4CI. Al2O~/ultrasound, then MeCOCI, DIPEA** *(55%),* **then 6N HCI, reflux** (99%)

Scheme 53

VIII. 2-AMINOBICY CLO[2.2.1]HEPTANE-2-CARBOXYLIC ACIDS

Two general methodologies were used for the preparation of heterosubstituted 2 **aminonorbornane-2-carboxylic** acid derivatives: *i)* the Diels-Alder reaction applied to heterosubstituted dienes or dienophyles (Meth. C) and *ii)* the **hetero-functionalization** of norbornane/norbornene ring (Meth. A) obtained through a Diels-Alder reaction. The first offers the advantages depicted for cyclohexyl derivative syntheses and is the more representative. The formation of both *exo-* and *endo-* diastereomeric compounds is achieved. As heteroatoms, only the fluoro, oxygen and sulfur atoms are linked to the norbornane ring.

1. Halo Ac~~~~~~~c Compounds

The diastereoselective synthesis of 2-fluoro-6-aminonorbornane-2,6-dicarboxylic acid⁸² was reported from the 2-fluoronorbornene derivative **exo-347a,** obtained **as** single diastereomer from **2-346a** by cycloaddition reaction with cyclopentadiene **1361** (Meth. **C)** (Scheme *54).*

a) cyclopentadiene **136l**, Et₂AlCl, CH₂Cl₂ (347a: 72%, 347b: 90%); b) BH₃: SMe₂, C₁₂H₂₆/THF, then H₂O₂, NaOH (68%); c) Dess-Martin reagent, CH₂Cl₂ (99%); d) (S)-1-phenylethylamine, TiCl₄, then TMSCN, AICI3 (70%); e) HCI, AcOH, then CHzN2 **(49%); f) Ti(O-ipr)4,** BnOH (69%); **g)** HzPd(0H) (68%); h) NaOH, then BnOH, EDC (70%)

Scheme 54

Compound **347a** was hydroxylated giving two regioisomeric compounds **348** and **349** (1.7: 1) with **ex0** selectivity. The major isomer **348** was oxidized to the corresponding ketone which reacted poorly in both the Strecker and Bucherer reactions. However, using (S)-phenylethylamine in the presence of TiCl₄ followed by trimethylsilyl cyanide, two diastereomers **350a**,b (3 1 :39) were formed. After separation, **350a** and **350b** were transformed into the corresponding chiral amino acids **352a,b** according **to** Scheme *54.* The same authors report on the chiral synthesis of **lS,2S,4S,6S-352a.** Starting from chiral acrylate Z-346b and cyclopentadiene **1361** the chiral **exo- 347b** (92% *d.e.)* was obtained and transformed into **(-)-347a.** According to the synthetic pathway depicted before, **(-)-347a** was transformed into the chiral compound **(-)-352a.**

2. Oxygen Ac12~z~llc Compounds

Both 3-, 5- and 6-hydroxy $Ac_[2,2,1]c$ compounds were prepared and, depending on the synthetic approach, the *exo* and/or *endo* selectivity was achieved. Regio- and diastereoselective syntheses of polyhydroxylated compounds **are** described. **The** Diels-Alder reaction of heterosub-

stituted dienophiles and cyclopentadiene is the classical method to obtain 2-amino-3-hydroxynorbornene-2-carboxylic acids (Meth. C).

Cufos at al.⁸³ reported the synthesis of a mixture of exo-355 and endo-355 adducts (4:1) by a "one pot" reaction starting from hippuric acid 353, trifluoroacetic anhydride and cyclopentadiene, via oxazolone intermediate 354 (Scheme 55). The hydrolysis of the oxazolone ring was

not reported. The 2-amino-3-hydroxynorbornane-2-carboxylic acids exo-360 and endo-360^{84a} (Scheme 56) were obtained from oxazolone Z-12i and cyclopentadiene 136j (Meth. C). The

a) cyclopentadiene 136j, Lewis acid, CH₂Cl₂ (from 12i: 70-90%; from 177: 45-50%); b) 357a: THF, HCl (100%); 358a: EtOH, Me₂NH (60%), 359a: EtOH, H⁺ (90%); c) H₂/ Pd, EtOH (100%); d) 20% HCl, 100°C (80-95%); e) 136, Mg(ClO₄)₂, ultrasound, CH₂Cl₂ (90%)

Scheme 56

cycloaddition reaction was catalyzed by Lewis acids and formed two cycloadducts exo-356 and endo-356 (70:30). Different Lewis acids (EtAlCL₂, Mg(ClO₄)₂, Ce(OTf)₄, and **Yb(OTf)**₄) were tested and the best results were obtained with EtAlCl, (74%) , Li(ClO_a), (88%) and Mg(ClO_a), (90%). The hydrolysis of the oxazolone ring to compounds **357a** followed by deprotection to compounds **358a**, reduction and hydrolysis gave the corresponding amino acids *exo-* and *endo-***360.** The synthesis of **1-amino-2-ethoxycarbonyloxyacrylate 177d** and of the chiral analogues **177e,f** was reported (Scheme 56).^{84b} They were used for a cycloaddition reaction with cyclopentadiene. The acrylate **177d was** less reactive than oxazolone **121** and the reaction worked both with EtAlCL, **(45%)** and Mg(ClO,), **(50%),** giving cycloadducts **exo-** and **endo-359b.** Interestingly, the *exo/endo* ratio was reverted (25:75) when starting from the same acrylate (177d) and EtAlCl₂. Reaction of chiral acrylates 177e,f using $Mg(CIO₄)$, and ultrasound resulted in an improvement of yields **(90%)** and a **7030** *exo/endo* ratio was obtained. The cycloaddition reaction gave four diastereomeric cycloadducts **exo-359cd,, exo-359'c,d, endo-359c,d** and **endo-359'c,d** with satisfactory diastereoselectivity **(em: 79% d.e.;** *endo:* **87%** *d.e.).*

The regioselective hydroboration of a series of norbomene amino acid derivatives **362** was studied by Brands⁸⁵ at al. using both BH₃.THF and catecholborane in the presence of a

c: $R = Me$; $R_1 = CHO$; $R_2 = CH(Me)NBn_2$ d: $R = Me$; $R_1 = H$; $R_2 = CH(Me)NBn_2$ **e**: $R = Me$; $R_1 = CO_2Me$ $R_2 = CH(Me)NBn_2$ **b**: $R = Et$; $R_1 = COPh$; $R_2 = H$

a) BH₃•THF, then H₂O₂, then Ac₂O, TEA, DMAP, CH₂Cl₂; b) catecholborane, (Cl(COD)₂Rh)₂•4Ph₃P, THF, then H_2O_2 , then Ac_2O , TEA, DMAP, CH_2Cl_2

Scheme 57

observed in all cases, but the regioselectivity at C-5 (compound **364)** and C-6 (compound **363)** was dependent on the substituents and kind of the transition metal catalyst. Starting from 362a and operating with BH, the **363/364** ratio is **62:38,** but in presence of the catalyst a inverted ratio was observed (22:78). Moreover, when the compounds **362c-d** functionalized at C-3 were used, the C-6 regioisomer **363** was preferred in all cases and its amount was **increased** by use of the catalyzed reaction.

Taking advantage on the use of the iodolactonization reaction starting from **endo**carboxylates (Meth. A), the regio and stereoselective functionalization of C-6 with an hydroxy group can be done giving the *N,O-trans* compounds **(Scheme** *58).*

a) **12,** NaHC03 **(4650%); b)** Zn, AcOH (40-92%); c) MeOH, **THF.** NaOH; d) NaHC03,Iz. KI, H20; e) SOC1₂, reflux, then NaN₃, acetone, then EtOH reflux; f) (nBu) ₃SnH, azobisisobutyronitrile, THF, Et₂O; g) I_2 , CH₂Cl₂ (93%) **Scheme 58**

When compounds **endo-365a** and **365b** were treated with iodine in basic conditions the iodo lactone derivatives **366a,b** were isolated, respectively. Attempts to eliminate the iodine atom with zinc in acetic acid produced the starting olefins 365 (Scheme 58).⁸⁶ Protected lactone 370 was prepared from **diethyl2,2-norbomenedicarboxylate 367a** (n = **1)** which was selectively hydrolyzed to **exo** carboxylate **367b** (Meth. A) (Scheme *58).* The iodolactonization reaction gave the iodo lactone **368** which was transformed, by **Curtius** reaction, into **369.** Iodine elimination with tributyltin hydride afforded lactone **370.8'** Instead, when iodo-oxazination reaction on *em* compounds was used, the functionalization of C-6 with the oxygen atom (Meth. A) (Scheme *58)* occurred giving the *N,O-cis* compound. From *exo* ester 372 and using iodine, the oxazine derivative 373 was obtained which was the precursor of compound 374 an inhibitor of protein kinase.⁸⁸ The direct formation of amino acid **endo-375,** having two *trans* hydroxy groups at C-3 and C-6, was observed when the *endo* ester **359a** was hydrolyzed in acidic conditions (Meth. A) (Scheme *59).89* Four diastereomeric 2-amino-3,5,6-trihydroxynorbornanecarboxylic acids⁸⁹ were prepared with *transcis* (compound **exo-380),** *cis-cis* (compound **endo-380),** *trans-trans* (compound **em-381)** and **cis***trans* (compound **endo-381)** relationships between OH-3 and OH-5 and between the latter and OH-6. They were obtained from esters **exo-** and **endo-359a** taking advantage of the presence of the double bond and the benzoylamino group in the **ex0** series, and of the carboxy group in the *endo* series (Meth. **A).** When esters *em-* and **endo-359** were reacted with osmium tetroxide the

Scheme 59

diols exo-376 and endo-376 were obtained, respectively, which were hydrolyzed to aminoacids exo- and endo-380. The epoxidation reaction of ester exo-359 produced compound exo-377 which was transformed into oxazine exo-378 and hydrolyzed to amino acid exo-381. From ester endo-359, the epoxide endo-377 was obtained together with the lactone endo-380. The epoxide endo-377 was transformed into endo-379 and hydrolyzed to amino acid endo-381.

3. Sulfur $Ac_{(2,2,1)}c$ Compounds

Three diastereomeric 2-amino-6-methylthio-norbornane-2-carboxylic acids were reported in the literature (Meth. A). The first one⁹⁰ (Scheme 60), with an exo methylthio group, was prepared from acetylthio-ketone 384 obtained from alcohol 382 through addition of thiolacetic acid followed by oxidation of the hydroxy group (regioisomeric mixture: 383/384, 1:1). Compound 384 was converted into the methylthio derivative and, by the Bucherer reaction, transformed into a mixture of diastereomeric spirohydantoins 385 and 386 (1:4). The hydrolysis of the major isomer 386 gave amino acid 387.

a) AcSH, C₆H₆, AlBN; **b**) Jones' reagent, $0^{\circ}C$ (88%); c) MeI, MeONa (85%); d) (NH₄)₂CO₃, KCN (95%); e) Ba(OH)₂ 120°C, then H⁺(90%)

Scheme 60

Both exo^{-91} and endo-amino acids⁹² 392 (Meth. A) (Scheme 61) having an endothiomethyl group were reported by Glass *at al.* starting in the first case from alcohol **390** obtained from olefin **388 as** a regioisomeric mixture with **389** (53, 83%). Compound **390 was**

a) BH3, THF, then H202, NaOH (83%); b) TBDMSCI, imidazole DMF; c) DIBALH, toluene; d) DMSO, oxalyl chloride, $CH_2Cl_2(28\%)$; e) $(NH_4)_2CO_3$, KCN, 80°C, then Ba(OH)₂, then H⁺(51%); **f) N,N-diisopropylamine, THF, -78T, BuLi, HMPA, 0-(mesitylenesulfonyI)hydroxylamine (38%); g) MeOH, H20, NaOH**

Scheme 61

transformed into ketone **391** in low yield **(28%)** which gave **392 (51%)** after Bucherer reaction **and** hydrolysis. **Endo-392** was obtained by a basic amination reaction of ester **393.** The reaction gave a single diastereomer **394** which was hydrolyzed to the corresponding **amino** acid *endo-***392.**

M. 2-AMINOBICY CLO[2.2.2]OCTANE-2-CARBOXYLIC ACIDS

Only oxygen substituted $Ac_{[2,2,1]}c$ compounds were described in the literature. The 1methoxy- or 1-silyloxy-1,3-cyclohexadienes 136m,n were used in the cycloaddition reaction with several dienophiles for the preparation of **2-aminobicyclo[2.2.2]oct-5-ene-2-carboxylic** acid derivatives in which an oxygen atom is linked at C-1 (Meth. C).

1. Oxygen $Ac_{(2,2,1)}c$ Compounds

Reacting diene 136m with acrylate 177c gave a single regioisomeric adduct exo-395 (Scheme 62).⁹³ Using chiral S-methyleneoxazinone 396 and diene 136m the chiral spiro compound 397 was obtained as the major isomer (¹H NMR: 397: 94%; other diastereomers: 6%) (Scheme 62).⁹⁴

The reaction of aminoacrylates 177g-l with diene 136n gave two diastereomeric compounds exo-398 and endo-398 (Scheme 63) with the same regiochemistry but with a

403b: $X = I$, $R = H$

a) PhMe, 120°C (51-72%); b) R = SiMe₃: MCPBA, NaHCO₃, toluene, 25°C (75-97%); c) CH₂Cl₂, TiCl₄, -78°C (36-66%); d) R = H: CH₂Cl₂, BF₃•Et₂O (100%), then MCPBA, toluene, 0°C (67%); e) NBS, dioxane, H₂O, 25°C (endo: 60%, exo: 88%); f) THF, tBuOK, -78°C (60%); g) I₂ CHCl₃, H₂O (lactone 74%, urethan 22%)

Scheme 63

CELMI AND POCAR

different *exo/endo* ratio depending on the substituent on nitrogen atom.95 In fact, from **177g** an 1.6: 1 *exdendo* ratio was found. In other cases the ratio was in favor of the **endo** isomer and the best result was obtained when $R_2 = C F_3 CO$ *(exo/endo,* 1:3.3). The adducts were epoxidized, the stereochemical results depending both on the *exo* or endo configuration of the starting material and, in the case of the *exo* adduct, on the kind of R_1 and R_2 groups.⁹⁶

Starting from the *endo* adducts **398,** the epoxidation reaction took place from the less hindered face of the olefin giving single diastereomers **anti-399** which were transformed into lactones **400** by reacting with TiCl,. Instead, starting from the **exo** adducts **398** a mixture of isomeric epoxides **anti401** (major) and **syn-401** was obtained. Interestingly, when **exo-398** having **R,=** Phth was desilylated and then epoxidized, only the *syn* epoxide **401** was otained. By contrast, starting from $exo-398$ with $R₂= CF₃CO$ the same reaction sequences produced only the **anti** epoxide **401.** In order to obtain the *syn* epoxides the same authors tested a different synthetic approach via halohydrin.⁹⁷ Treating the *endo* adduct 398 with NBS in aqueous dioxane, a mixture of the corresponding halohydrin **402** and lactone **403a** was obtained **(402/403a,** *5:* 1). The halohydrin **402 was** quantitatively transformed into **403a** by a basic treatment. Under the same reaction conditions the **exo** adduct **398** afforded a mixture of bromo oxazine **404** and bromo urethan **405a (404/405a,** 1:1.4). By reaction with iodine in water and starting from a mixture of *exo* and **endo** adducts the corresponding iodo lactone **403b** and iodo urethan **405b** were obtained. The 6-hydroxy lactone **371** was obtained from **367a** (n = 2) according to the synthetic pathway depicted for the norbornane analogue **370** (Meth. A) *(Scheme* **58).87**

X. MISCELLANEOUS

The addition of amines to 1,2-epoxyindane- 1-carboxylate derivatives **406** gave a mixture of regioisomeric β -amino- α -hydroxy- and α -amino- β -hydroxy acid derivatives **407** and **408** of the ester function in 406 and of the attacking nucleophile.⁹⁸ on of amines to 1,2-epoxyindane-1-carboxylate derivatives 406 g

eric β-amino-α-hydroxy- and α-amino-β-hydroxy acid derivative

(b). A different ratio of the two compounds was obtained depending

function in 406 and of t

A series of ethyl **2-aminoindane-2-carboxylates 410** was prepared as precursors for the preparation of compounds having anthropodicide activity *(Scheme 65).* Keto esters **409** were aminated in basic conditions using the imine derived from cyclohexanone and hydroxylamine-0 sulfonic acid giving **41O.'Oo** Steroidal amino acid **415** (having digitalis steroidal skeleton), with

the amino acid functions on C-5 ring, was produced from steroidal compound **411** which was transformed into acyl aide **412** and pyrolyzed to the cyclic urethan **413.** Hydrolysis of **this** ring and protection both of nitrogen atom and tertiary hydroxy produced intermediate **414** which was transformed into amino acid **415** (Scheme **66).99**

a) triphosgene, pyridine, CH2C12; **b)** NaN3. Me2CO (72%); c) tefrachloroethane, 140°C (70%); d)KOH, EtOH; e) DHP, TsOH, dioxane; f) chloromethyl ethyl ether, DIPEA, CH₂Cl₂, 0°C; g) Boc₂O, NaOH, $tBuOH (52\%)$; h) PDC, CH₂Cl₂; i) KMnO₄, NaH₂PO₄ $tBuOH$; 1) TsOH, MeCN/H₂O (48%)

Scheme 66

Carbomethoxyaziridines **417,** obtained from **416** and phenyl azide, were the starting material for the diastereoselective preparation of tetrahydronaphthalene derivatives **418** *(85-* 90%) after their treatment with organic acids (Scheme 67).¹⁰¹

a) PhN3, CH2C12; b) RC02H, benzene *(8590%)*

Scheme 67

ABBREVIATIONS

 A **IBN** = azobisisobutyronitrile

 $Boc-L-Phe = N-t-butoxycarbonyl-L-phenylalanine$

 $Bn = benzyl$

 $Bz =$ benzoyl

 $CAN = ceric$ ammonium nitrate

 $Cbz = carbobenzyloxy$

 $CIPBA = chloroperbenzoic acid$

 $CTACI = cetyltrimethylammonium chloride$

 $Cu(TBS)$ ₂ = bis(N-t-butylsalicylaldimine) copper (II)

 $DABCO = 1.4$ -diazabicyclo[2.2.2] octane

DAST = diethylaminosulfur trifluoride

 $DBN = 1.5$ -diazabicyclo $[4.3.0]$ non-5-ene

 $DBU = 1,8$ -diazabicyclo[5.4.0]undec-7-ene

 $DEC = 1-(3-dimethylaminopropyl)-3-ethylcarbodimide$

 $DEAD =$ diethyl azodicarboxylate

 $DHP = 3.4$ -dihydro-2H-pyran

 $(DHOD)$ ₂ $PHAL = bis(dihydroquinidine)$ phthalazine

 $DIBAL = disobutylaluminum hydride$

 $DIPEA = disopropylethylamine$

 $DMAP = 4$ -dimethylaminopyridine

 $DME = ethylene$ glycol dimethyl ether

 $DMF = dimethylformamide$

 $DPM = diphenylmethyl$

 $DPPA = diphenylphosphoryl azide$

 $EDC = 1-ethyl-3-(3-dimethylaminopropyl) carbodimide$

 $HMPA = hexamethylphosphorotriamide$

 $Ipc = isopinocamphenyl$

 $LDA =$ lithium diisopropylamide

 $LHMDS = lithium hexamethyldisilylamide$

 $MCPBA = m$ -chloroperbenzoic acid

 $Men = menthyl$

 $MsCl = mesyl$ chloride

 $NHPT = 1,2$ -dihydro-1-hydroxypyridine-2-thione

 $NBS = N-bromosuccinimide$

 $NIS = N$ -iodosuccinimide

 $NMM = N$ -methylmorpholine

 $NMO = N$ -methylmorpholine N-oxide

NMP = **1-methyl-2-pyrrolidinone**

Ns = 4-nitrobenzenesulfonyl

PCC = pyridinium chlorochromate

PDC = pyridinium dichromate

PEA = phenylethylamine

Piv = pivaloyl

 $PMB = p$ -methoxybenzyl

 $PMBNH₂ = p$ -methoxybenzylamine

PS = Pseudomonas cepacia (Amano)

PTC = phase-transfer catalysis

Py = pyridine

 $RCM = ring closing metathesis$

 $TBDMSCl = t$ -butyldimethylsilyl chloride

TBHP = r-butylhydroperoxide

TEA = triethylamine

TFA = trifluoroacetic acid

Tf = trifluoromethylsulfonyl

 $TIMEDA = N, N, N', N'$, -tetramethylethylenediamine

TMGA = tetramethylguanidine

 $TMSOTf =$ trimethylsilyl triflate

TMSCl = chlorotrimethylsilane

 $TMSCN = trimethylsylil cyanide$

 $TMS =$ trimethylsilyl

WSCD = 1-ethyl-3-(3-dimethylaminopropyl)carbodiimide HCl

REFERENCES

- **1.** (a) C. H. Stammer, *Terrahedron,* **46,2231 (1990);** (b) K. Burgess, K.-K. Ho and D. Moye-Sherman, *Synlerr.,* **575, (1994);** (c) C. Cativiela and M. D. Diaz-de-Villegas, *Tetrahedron: Asymmetry,* **11,645 (2000).**
- **2.** (a) F. Tellier, F. Acher, I. Brabet, J.-P. **Pin** and R. Azerad, *Biorg. Med. Chem.,* **6, 195 (1998);** (b) A. B. Charette and B. Cod, J. *Am. Chem.* **SOC., 117, 12721 (1995);** (c) E. Buñuel, C.; Cativiela and M. D. Díaz-de-Villegas, *Tetrahedron: Asymmetry*, 7, 1521 **(1996);** (d) H. Mickos, K. Sundberg and B. Luning, *Acra Chim. Scad.,* **46,989 (1992).**
- **3.** (a) C. Toniolo, M. Crisma, *G.* Valle, G. M. Bonora, B. Barone, E. Benedetti, B. D. Blasio, V. Pavone, C. Pedone, A. Santini and F. Lelj, *Pept. Chem.*, 45, (1987); (b) K. Burgess, K.-K. Ho and B. M. Pettitt, J. *Am. Chem. SOC.,* **116,799 (1994);** (c)) K. Burgess, K.-K. Ho and B. M. Pettit, J. *Am. Chem.* **SOC., 117,54 (1995);** (d) K. Burgess, K.-K. Ho and B. Pal, J. *Am. Chem.* **SOC., 117,3808 (1995);** (e) A. I. Jimertez, C. Cativiela, A. Aubry and M. Marraud, J. *Am. Chem.* **SOC., 120,9452 (1998);** *(f)* M. Llinas-Brunet, M. D. Bailey, D. Cameron, E. Ghiro, N. Goudreau, M.-A. Poupart, **J.** Rancourt and Y. **S.** Tsantrizos, PTC Int. Appl. WO 00 09,558; *Chem. Abst.*, 132, 175808 (2000).
- **4.** M. J. Sloan and K. L. Kirk, *Tetrahedron Lea,* **38, 1677 (1997).**
- **5.** M. Kirihara, T. Takuwa, M. Kawasaki, H. Kakuda, **S.** Hirokami and H. Takahata, *Chem***isrry** *Lerr.,* **405, (1999).**
- **6.** (a) J. Bland, C. H. Stammer and K. I. Varughese, J. *Org. Chem.,* **49, 1634 (1984);** (b) J. Bland, A. Shah, A. Bortolussi and C. H. Stammer, J. *Org. Chem.,* **53,992 (1988);** (c) L. Wick, C. Tamm, M. Neuburger and M. Zehnder, *Tetrahedron,* **51,10219 (1995);** (d) F. Clerici, M. L. Gelmi and A. Gambini, J. *Org. Chem.* **64,5764 (1999).**
- **7.** K. Li, W. Du, N. L. **S.** Que and H. Liu, J. *Am. Chem. SOC.,* **118,8763 (1996).**
- **8.** L. Wick, C. Tamm and T. Boller, *Helv. Chim. Acra,* **78,403 (1995).**
- **9.** A. L. Schwas and J. Warkentin, *Can.* J. Chem., *66,* **1686 (1988).**
- 10. R. Badorrey, C. Cativiela, M. D. Díaz-de-Villegas and J. A. Gálvez, *Tetrahedron: Asymmerry,* **11,1015 (2000).**
- 1 1. F. Clerici, M. L. Gelmi and D. Pocar, J. *Org. Chem.,* **64,726 (1999).**
- **12.** F. Clerici, M. L. Gelmi, D. Pocar and T. Pilati, *Tetrahedron: Asymmerry,* **12,2663, (2001).**
- 13. T. M. Shoup and M. M. Goodman, *J. Labelled Compd. Radiopharm.*, **42**, 215 (1999).
- **14.** M. M. Goodman and T. M. Shoup, PCT Int. Appl. WO **97 17,092 (1997).** *Chem. Absr.* **127,51001 (1997).**
- 15. L. Littman, C. Tokar, S.Venkatraman, R. J. Roon, J. F. Koemer, M. B. Robinson and R. L. Johnson, J. *Med. Chem,* 42,1639 (1999).
- 16. R. D. Allan, J. R. Hanrahan, T. W. Hambley, G. A. R. Johnston, K. N. Mewett and A. D. Mitrovic, J. *Med. Chem.,* 33,2905 (1990).
- 17. H. Dworschak and F. Weygand, *Chem. Ber.,* 101,302 (1968).
- 18. Y. Gaoni, *Org. Prep. Proc. Int.*, 27, 185 (1995) and references therein.
- 19. J. R. Hanrahan, P. C. Taylor and W. Errington, J. Chem *Soc., Perkin Trans. I,* 439 (1997).
- 20. R. S. Atkinson, E. Barker, P. J. Edwards and G. A. Thomson, *J.* Chem. *Soc., Perkin Trans. I,* 1533 (1995).
- 21. B. F. Borrow, S. D. Mills and W. B. Turner, *Chem.* Commun.,75 (1965).
- 22. B. F. Borrow, and W. B. Turner, Brit. 1,043,508 (1966); *Chem. Abst.,* 65,2029e (1966).
- 23. J. D. Huddle and C. G. Skinner, J. *Med. Chem.,* 14,545 (1971).
- 24. D. E. Gaitanopoulos, J. Nash, D. L. Dunn and C. G. Skinner, *J. Med. Chem.*, 19, 342 (1976).
- 25. Y. Ohfune, K. Nanba, I. Takada, T. Kan, M. Horikawa and T. Nakajima, *Chirality,* 9,459 (1997).
- 26. K. Hammer and K. Undheim, *Tetrahedron,* 53,5925 (1997).
- 27. D. Ma, J. Ma and L. Dai, *Tetrahedron: Asymmetry,* 8,825 (1997).
- 28. D. M. Hodgson, A. J. Thompson, S. Wadman and C. J. Keat, *Tetrahedron,* 99,10815 (1999).
- 29. K. Hammer, J. Wang, M. L. Falck-Pedersen, C. Romming and K. Undheim, J. *Chem. Soc. Perkin I,* 1691 *(2OOO).*
- 30. K. F. Mcclure, R. P. Robinson, EP 1041072 (2000); *Chem. Abstr.,* 133,266834 (2000).
- 31. (a) A. Hui, A. J. Fairbanks, R. J. Nash, P. M. de Q. Lilley, R. Storer, D. J. Watkin and G. W. J. Fleet, *Tetrahedron* Lett., 35,8895 (1994); (b) A. J. Fairbanks, A. Hui, B. M. Skead, P. de Q. Lilley, R. B. Lamont, R. Storer, J. Saunders, D. J. Watkin and G. W. J. Fleet, *Tetrahedron Len.,* 35,8891 (1994).
- 32. A. J. Fairbanks, R. **P.** Elliott, C. **Smith,** A. Hui, G. Way, R. Storer, H. Taylor, D. J. Watkin, B. G. Winchester and G. W. J. Fleet, *Tetrahedron Lett.,* 34,7953 (1993).
- 33. M. Vincent, G. Remond and J. Bure, *Eur. Pat. Appl.* 1,139 (1997). *Chem. Abst.,* 92, 163619~ (1980).
- 34. A. W. Coulter, J. B. Lombardini, J. R. Sufrin and P. Talalay, Mol. Pharmacol., 10, 319 (1974).
- 35. L. Amori, G. Costantino, M. Marinozzi, R. Pellicciari, F. Gasparini, J. P. Flor, R. Kuhn and I. Vranesic, *Bioorg. Med. Chem. Lett.,* 10, 1447, (2000).
- 36. S. L. Crooks, M. B. Robinson, J. F. Koemer and R. L. Johnson, J. *Med. Chem.,* 29, 1988 (1986).
- 37. C. R. Johnson, B. R. Boettcher, R. E. Cherpeck and M. G. Dolson, *Bioorg. Chem.,* 18,154 (1990).
- 38. F. Clerici, M. L. Gelmi and A. Gambini, J. *Org. Chem.* 64,5764 (1999).
- 39. A. Avenoza, C. Cativiela, M. A. Femandez-Recio and J. M. Peregrina, *Tetrahedron: Asym*metry, 10, 3999 (1999).
- 40. M. Schlauch, F.-J. Vok, K. P. Fondekar, **J.** Wede and A. W. Fraham, J. *Chromutogr. A,* 897, 145 (2000).
- 41. F. Roberts and P. V. Tabernier, *Br.* J. Pharmucol., 61,476P (1977).
- 42. H. N. Christensen and A. M. Cullen, *Biochem. BiophyS. Acra,* 298,932 (1973).
- 43. T. S. Yokum, M. G. Bursavich, *S.* A. Piha-Paul, D. A. Hall, M. L. Mclaughlin, *Tetrahedron* Lett., 23, 4013 (1997).
- 44. F. Himmelsbach, G. Linz, H. Pieper, V. Austel, T. Miieller and J. Weisenberger, DE 4326344 (1995). *Chem. Abstr.,* 123,256522 (1995).
- 45. J. R. **Piper,** C. R. Shingfellow and T. P. Johnson, *J. Med. Chem.,* 9,911 (1966).
- 46. J. F. W. Keana, J. S. Bland, P. E. Eckler, V. Nelson and J. Z. Gougoutas, J. *Org. Chem.,* 41, 2124 (1976).
- 47. G. A. Kraus, George and F. X. Yu, *Synth. Commun.,* 19,2401 (1989).
- 48. F. Clerici, M. L. Gelmi, A. Gambini and D. Nava, *Tetrahedron.,* 57, 6429 (2001).
- 49. A. Avenoza, J. I. Barriobero, C. Cativiela, M. A. Fernández-Recio, J. M. Peregrina and F. Rodriguez, *Tetrahedron,* 57,2745 (2001).
- 50. A. Avenoza, J. H. Busto, C. Cativiela and J. M. Peregrina, *Amino Acids,* 18, 117 (2000).
- 51. R. M. Ortuño, J. Ibarzo, J. D'Angelo, F. Dumas, A. Alvarez-Larena, and J. F. Piniella, *Tetrahedron: Asymmetry,* 7, 127 (1996).
- 52. S. G. Pyne, J. Safaei-G, D. C. R. Hockless, B. W. Skelton, A. N. Sobolev and A. H. White, *Tetrahedron,* 50,941 (1994).
- 53. A. Avenoza, J. H. Busto, C. Cativiela and J. M. Peregrina, *Tetrahedron, 50,* 12989 (1994).
- 54. E. Buñuel, C. Cativiela and M. D. Diaz-de-Villegas, *Tetrahedron: Asymmetry*, 7, 1431 (1996).
- 55. A. Avenoza, J. H. Busto, M. París, C. Cativiela and J. M. Peregrina, *J. Heterocyclic Chem.*, *34,* 1099 (1997).
- 56. C. Rodriguez-Garcia, J. Ibarzo, A. Alvarez-Larena, V. Branchadell,A. Oliva and R. M. Ortuiio, *Tetrahedron,* 57, 1025 (2001).
- 57. A. Avenoza, C. Cativiela, M. A. Fernández-Recio and J. M. Peregrina, J. Chem. Soc. *Perkin I,* 3375 **(1999).**
- 58. H. E. Album and W. Dvonch, U. S. 3,592,812 (1971); *Chem Abst.,* 75,76813m (1971).
- 59. K. P. Fondekar,F.-J. **Volk** and A. W. Frahm, *Tetrahedron: Asymmetry,* 10,727 (1999).
- 60. E. Felice, S. Fioravanti, L. Pellacani and P. A. Tardella, *Tetrahedron Lett.,* **40.44** 13 (1999).
- 61. S. Krikstolaitytb, K. Hammer and K. Undheim, *Tetrahedron Lett.,* 39,7595 (1998).
- 62. K. Hammer, C. Rgmming and K. Undheim, *Tetrahedron,* 54,10837 (1998).
- 63. S. Krikstolaitytè, A. Sackus, C. Romming and K. Undheim, *Tetrahedron: Asymmetry*, 12, 393 (2001).
- *64.* A. Avenoza, C. Cativiela and J.M. Peregrina, *Tetrahedron, 50,* 10021 (1994).
- 65. A. Avenoza, C. Cativiela, M. A. Fernánder-Recio and J. M. Peregrina, Tetrahedron: Asym*metry,* 7,721 (1996).
- 66. A. Avenoza, C. Cativiela, M. Paris, J. Peregrina and B. Saenz-Torre, *Tetrahedron: Asymmerry,* 8, 1123 (1997).
- 67. Y. Maki and T. Masugi, *JP* 48067254 (1073). *Chem. Abstr.,* 80,3169 (1974).
- 68. **A.** M. M. Majalli, WO 0043352 **(2000).** *Chem Abstr.,* 133,135607 **(2000).**
- 69. A. Britten, G. Lockwood, J. *Chem.* **SOC.** *Perkin Trans.* 1,1824 (1973).
- 70. G. M. Ksander, P. J. Kukkola, L. A. Robinson, WO 9633170 (1996). *Chem Abstr.,* **126,** 31652 (1997).
- 71. R. Fischer, G. Beck, EP 595130 (1994). *Chem. Abstr.,* 121,255419 (1994).
- 72. S. Zen, Y. Takeda, A. Yasuda and S. Umezawa, Bull. *Chem.* **SOC.** *Jpn.,* 40,431 (1967).
- 73. K. Achiwa and S. Yamada, Chem *Pharm.* Bull. 14,537 (1966).
- 74. E. W. Logusch, D. M. Walker, J. F. McDonald, G. C. Leo and J. E. Franz, J. *Org. Chem.,* 53,4069 (1988).
- 75. D. S. Black, D. C. Craig, G. L. Edward and S. M. Laaman, *Biorg. Chem.,* 27,91 (1999).
- 76. **A.** Nakazato, T. Kumagai, K. Sakagami, R. Yoshikawa, Y. **Suzuki,** S. Chaki, H. Ito, T. Taguchi, S. Nakanishi and S. Okuyama, J. *Med. Chem.,* 43,4893 (2000).
- 77. *S.* M. Massey, J. A. Mom and M. J. Valli, EP 878463 (1998); *Chem. Abstr.* 130,4081 (1998).
- 78. *G.* Adam, P. N. Huguenin-Virchaux, V. Mutel, H. Stadler and T. J. Woltering, DE 19941675 (2000); *Chem. Abst.*, **132**, 166520 (2000).
- 79. J. A. Mom and M. J. Valli, EP 1000927 (2000); *Chem. Absf.,* 132,334795 (2000).
- 80. C. Dominguez, J. Ezquerra, S. R. Baker, S. Borrelly, L. Prieto, C. M. Espada and C. Pedregal, *Tetrahedron Lett.,* 39,9305 (1998).
- 81. C. Dominguez-Fernandez, J. **A.** Monn and M. J. Valli, EP 774454 (1997); *Chem. Absf.,* 127,66215 (1997).
- 82. H. Ito, A. Saito, A Kakuuchi and T. Taguchi, *Tetrahedron,* 55, 12741 (1999).
- 83. T. Cufos, P. Diaz and L. Stella, *Synlef?.,* 101 (1995).
- 84. (a) F. Clerici, M. L. Gelmi and A. Gambini, J. Org. *Chem.,* 65,6138 (2000); (b) G. Abbiati, F. Clerici, M. L. Gelmi, **A.** Gambini and T. Pilati, *J. Org. Chern.,* 66,6299 (2001).
- 85. K. M. J. Brands and A. S. Kende, *Tetrahedron Lett.*, 33, 5887 (1992).
- 86. C. Cativiela, M. D. Diaz de Villegas and J. A. Mayoral, *Tetrahedron,* 49,677, (1993).
- 87. J. Katsube; H. Shimomura, S. Inokuma and A. Sugie, US 4347254 (1982); *Chem. Absf.,* 98, 71905 (1983).
- 88. C. T. Iwasaki, H. Yamazaki, T. Nishitani, K. Kondo and T. Sato, *Chem. Pharm.* Bull., **40,** 122 (1992).
- 89. F. Clerici, M. L. Gelmi and **A.** Gambini, J. *Org. Chem.,* 66,4941 (2001).
- 90. G. Fantin, M. Fogagnolo, R. Guerrini, M. Marastoni, A. Medici and P. Pedrini, Tetrahe*dron,* 50, 12973 (1994).
- **91.** R. Glass, M. Sabahi and W. P. Singh, J. Org. *Chem.,* 57,2683 (1 992).
- 92. R. **S.** Glass, M. Hojatie, M. Sabahi, L. K. Steffen and G. *S.* Wilson, *J. Org. Chem.* 55,3797 (1990).
- 93. M. Souchet, J. Guilhem and F. **Le** Goffic, *Tetrahedron Lett.,* 28,2371 (1987).
- 94. R. Chinchilla, L. R. Falvello, N. Galindo and C. Najera, J. *Org. Chem.* 65,3034 (2000).
- 95. M. J. Crossley and A. W. Stamford, *Australian* J. *Chem.,* 47,1695 (1994).
- 96. M. J. Crossley and A. W. Stamford, *Australian* J. *Chem.,* 47, 1713 (1994).
- 97. M. J. Crossley, **S.** R. Davies, T. W. Hambley and Trevor W. *Australian* J. *Chem.,* 47,2221 (1994).
- 98. J. C. Craig, A. Dinner and P. J. Mulligan, J. *Org. Chem..* 39, 1669 (1974).
- 99. G. Fedrizzi, L. Bernardi, G. Marazzi, P. Melloni and M. Frigerio, J. *Chem.* **Soc.** *Perkin Trans. I,* 1755 (1995).
- 100. G. D. **Annis,** W. E. Barnette, **S.** F. McCann, and K. D. Wing, WO9211249 (1992); *Chem. Abst.,* 117,234026 (1992).
- 101. K. Tshiamala, J. Vebrel, B. Laude and R. Mercier, *Bull.* **SOC.** *Chim. Fr.,* 127,584 (1990).

(Received January 8,2002; in final form September 23,2002)